

- i) Suggest a suitable substance which could be used for the electrodes.

_____ 1mark

- ii) State the name of the gas given off at the electrode B?

_____ 1mark

- b) Write the reaction that takes place at each electrode when copper sulphate solution is electrolysed using :

i) Graphite electrodes

ii) Copper electrodes.

(4marks)

Electrode	Solution	Reaction takes place at the anode	Reaction takes place at the cathode.
Graphite electrode	Copper sulphate solution		
Copper electrode	Copper sulphate solution		

- c) What mass of metallic silver will be deposited when a current of 0.53A is passed through a silver nitrate solution for 95 minutes? (mass Ag = 108, N = 14, O = 16)
(4marks)

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Yi-Tong Ma



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