SS1 CHEMISTRY EXAM QUESTIONS FOR SECOND TERM PAPER 1: TEST OF OBJECTIVE QUESTIONS

- If the valency of hydrogen is one and that of oxygen is two the formula is
 (a)HO₂ (b) H₂O (c) 2OH (d) H₂O₃
- The IUPAC name for the compound ClCH2CH2OH is
 - (a) 1-chloropropanol (b) 1-chloropropanol (c) 1-chloroethanol (d) 2-chloroethanol
- Petroleum is a non-renewable source of energy because it
 (a) is formed naturally (b) is cheap (c) can be recycled after use (d) cannot be regenerated once used up
- One of the Isotopes of hydrogen is (a) Ozone (b) Deutenum (c) diamond
 (d) graphite
- 5. Which of the following arrangements shows increasing order of reactivity of the halogens?
 - (a) F₂> Cl₂> Br₂> I₂ (b) I₂ < Br₂< F₂ (c) F₂ < Cl₃ < Br₂ < I₂ (d)I₂ > Br₃ > Cl₂ > F₂
- Hydrogen is prepared in the laboratory by the action of dilute H2SO4 on the following metals except (a) Zn (b) Na (c)Al (d)Fe
- The gas given out when sodium reacts with water is (a) oxygen (b) hydrogen (c) nitrogen (d)Chlorine
- The minimum amount of energy required for effective collisions between reacting particles is known as (a) activation energy (b) bond energy (c) kinetic energy (d) potential energy
- On a large scale, hydrogen is produced from (a) producer gas (b) water gas (c) coke (d) coal tar
- In which group is oxygen in the periodic table? (a) IV (b) VI (c) VII (d) IV
- The bond formed between H2O and H+ to form hydroxonium ion H3O+ is
 (a) dative b) covalent (c) electrovalent (d) sodium
- Which of the following is an allotrope of oxygen?(a) oxide (b) Ozone (c)dismond(d)graphite
- Oxygen is prepared by heating. (a) Na₂CO₃ (b) KClO₃(c) K₂SO₄ (d) H₂S

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Jessica J Manson

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