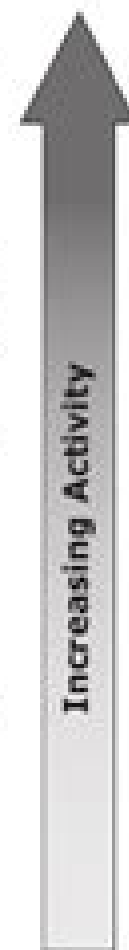


STAAR CHEMISTRY REFERENCE MATERIALS

POLYATOMIC IONS		SOLUBILITY OF COMMON IONIC COMPOUNDS IN WATER		ACTIVITY SERIES
Acetate	$\text{C}_2\text{H}_3\text{O}_2^-$, CH_3COO^-	<u>Soluble compounds contain</u>	<u>Common exceptions</u>	<u>Metal</u>
Ammonium	NH_4^+	$\text{C}_2\text{H}_3\text{O}_2^-$, CH_3COO^-	None	Lithium
Carbonate	CO_3^{2-}	NH_4^+	None	Potassium
Chlorate	ClO_3^-	NO_3^-	None	Barium
Chlorite	ClO_2^-	CN^-	None	Calcium
Chromate	CrO_4^{2-}	ClO^-	None	Sodium
Cyanide	CN^-	CO_3^{2-}	None	Magnesium
Dichromate	$\text{Cr}_2\text{O}_7^{2-}$	ClO_2^-	None	Aluminum
Hydrogen carbonate	HCO_3^-	ClO_3^-	None	Manganese
Hydroxide	OH^-	ClO_4^-	None	Zinc
Hypochlorite	ClO^-	Br^-	Compounds of Ag^+ , Pb^{2+} , and Hg_2^{2+}	Chromium
Nitrate	NO_3^-	Cl^-	Compounds of Ag^+ , Pb^{2+} , and Hg_2^{2+}	Iron
Nitrite	NO_2^-	I^-	Compounds of Ag^+ , Pb^{2+} , and Hg_2^{2+}	Cobalt
Perchlorate	ClO_4^-	SO_4^{2-}	Compounds of Sr^{2+} , Ba^{2+} , Pb^{2+} , and Hg_2^{2+}	Nickel
Permanganate	MnO_4^-	<u>Insoluble compounds contain</u>	<u>Common exceptions</u>	Tin
Phosphate	PO_4^{3-}	CO_3^{2-}	Compounds of NH_4^+ and the alkali metal cations	Lead
Sulfate	SO_4^{2-}	PO_3^{3-}	Compounds of NH_4^+ and the alkali metal cations	(Hydrogen)
Sulfite	SO_3^{2-}	CrO_4^{2-}	Compounds of NH_4^+ and the alkali metal cations	Copper
		$\text{Cr}_2\text{O}_7^{2-}$	Compounds of NH_4^+ and the alkali metal cations	Mercury
		OH^-	Compounds of NH_4^+ , the alkali metal cations, Ca^{2+} , Sr^{2+} , and Ba^{2+}	Silver
		S^{2-}	Compounds of NH_4^+ , the alkali metal cations, Ca^{2+} , Sr^{2+} , and Ba^{2+}	Platinum
				Gold



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Ying Liu

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