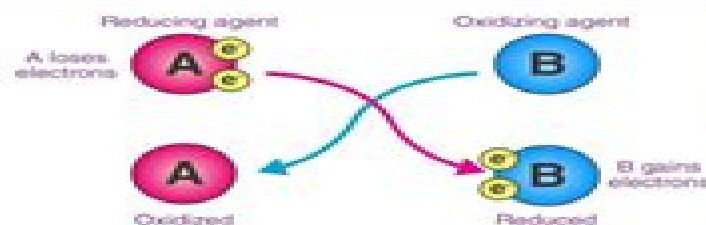


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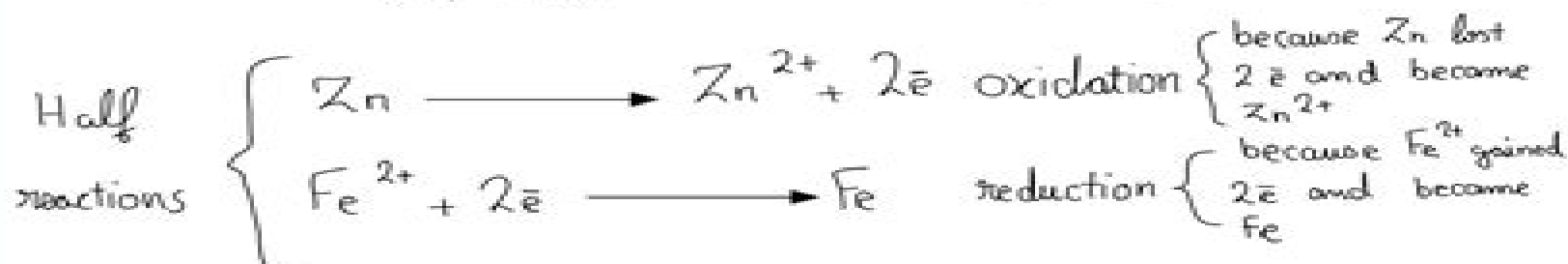
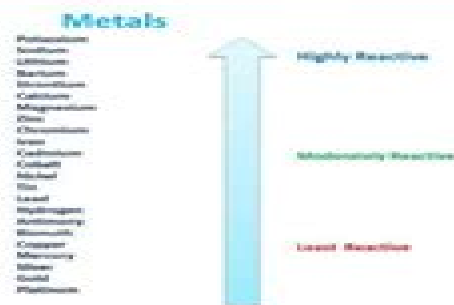
- A redox reaction is a chemical reaction where **both** oxidation and reduction take place
- An ion with positive charge is called Cations
- An ion with negative charge is called Anions
- Oxidation is a loss of electrons
- Reduction is a gain of electrons
- Not all the oxidation reactions involve oxygen

* The ion or molecule that accepts electrons is called the oxidizing agent - by accepting electrons it oxidizes other species.

* The ion or molecule that donates electrons is called the reducing agent - by giving electrons it reduced the other species.



- If a metal is more reactive than the metal corresponding to the ion there is going to be a reaction
- If a metal is less reactive than the metal corresponding to the ion there is **NOT** going to be a reaction



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