

CHAPTER 9 STUDY GUIDE

Chemical Reactions

Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write *yes* if evidence of a chemical reaction is present. Write *no* if there is no evidence of a chemical reaction.

- _____ 1. A tomato smells rotten.
- _____ 2. A drinking glass breaks into smaller pieces.
- _____ 3. A piece of ice melts.
- _____ 4. Drain cleaner is mixed with water and the solution becomes warm.
- _____ 5. Candle wax burns.
- _____ 6. Molten candle wax solidifies.
- _____ 7. Green leaves turn yellow and red as the seasons change.
- _____ 8. Baking powder produces a gas that makes a cake rise.

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

arrow	plus sign	(s)	(l)
reactant	product	(g)	(aq)

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is a(n) **(9)** _____, oxygen is a(n) **(10)** _____, and water vapor is a(n) **(11)** _____. In a chemical equation for this reaction, a(n) **(12)** _____ is used to separate hydrogen and oxygen from water vapor and energy. A(n) **(13)** _____ is used to separate the symbols for hydrogen and oxygen. A(n) **(14)** _____ symbol is used to tell the state of hydrogen in the reaction, a(n) **(15)** _____ symbol is used for the state of oxygen, and a(n) **(16)** _____ symbol is used for the state of water vapor.

Study Guide Section 2 Chemical Reactions

Arshad Iqbal



Study Guide Section 2 Chemical Reactions:

Student Notebook and Study Guide to Accompany The Human Body Bruce Wingerd, Patty Bostwick Taylor, 2013-02-01

This Student Notebook and Study Guide the ideal companion to Bruce Wingerd's *The Human Body* reinvents the traditional study guide by giving students a tool to help grasp information in class and reinforce learning outside of class Too often students struggle to both learn the concepts presented and simultaneously record crucial information The Student Notebook and Study Guide provides a structure for recording in class material that parallels the text's concept presentation and includes supplemental questions and activities for assignment outside of the classroom A complete answer guide for both the in class and out of class materials is available online

Organic Chemistry Principles and Industrial Practice Mark M. Green, Harold A. Wittcoff, 2003-09-19 Nylon plexiglas epoxy resin and Elmer's glue dynamite rubber tires and spandex These are a few among the multitude of industrial products produced using the principles of organic chemistry principles that are often taught to students without reference to the commercial and practical importance of the subject The marvelous theoretical principles on which organic chemistry is based are therefore often not fully appreciated by students of this subject Organic chemistry can appear dry meaningless and seemingly irrelevant In this textbook designed to be used in conjunction with classic texts of organic chemistry at the undergraduate level or standing alone for more advanced students two experts M M Green and H A Wittcoff bring the principles and the practice together Written for students and also giving much information that could be used to enhance teaching of the subject the book presented in ten concise chapters combines important industrial processes with the principles of organic chemistry The result is a source of otherwise barely accessible information In addition personal anecdotes from the authors vast experience make this a fascinating and indispensable textbook for everyone wishing to enhance the appreciation of this subject I have never come across such an enticing mix of stories of discovery with basic chemistry Roald Hoffmann Cornell University Simply put this book is a gem The chemistry described is rigorous but the warm humorous and conversational writing style makes the book a joy to read Dasan M Thamattoor Colby College This is a unique fascinating book that bridges organic chemistry principles with chemical industrial applications The story telling style make the reading learning experience extremely enjoyable Qiao Sheng Hu College of Staten Island City University of New York

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enables students with varied backgrounds to proceed together to significant levels of achievement

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Resources in Education, 2001-10

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