Name:

Garnet Lee

Date

3/23/21

Make a copy of this to your Google drive. Click on the Gizmo icon in Classlink, select the Solubility and Temperature Gizmo, and complete the following. Take a screenshot of your quiz results and attach to the bottom of this page.

Student Exploration: Solubility and Temperature

50 points

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: concentration, dissolve, homogeneous mixture, solubility, solubility curve, solute, solution, solvent

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

 What happens when you stir a spoonful of sugar into hot water? The sugar will dissolve

When sugar or another substance is dissolved in water, it disappears from view and forms a homogeneous mixture with the water, also called a solution.

If you can't see the sugar, how can you tell that it is there?

the color of the water changes and it will taste different

3. Does sugar dissolve more easily in hot water or cold water?

sugar dissolves more easily in hot water

Gizmo Warm-up

A solution generally consists of two parts, a **solute** that is dissolved and a **solvent** that the solute is dissolved into. For example, sugar is a solute that is dissolved into the solvent water. In the *Solubility and Temperature* Gizmo, you will study how temperature affects how much solute will dissolve in a solution.

To begin, check that Potassium nitrate is selected and the Temp. of the water is 20 °C. Click OK.

In this solution, what is the solute?

potassium nitrate What is the solvent?

water

Proteinsium witrete

ALCOHOLD TO BE

Add 5 g

App 20 a

Click Add 10 g to mix 10 g of potassium nitrate into the water.

<u>Teacher Answer Key Explore Learning Gizmo Solubility</u> <u>And Temperature</u>

J Rink

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Table of Contents Teacher Answer Key Explore Learning Gizmo Solubility And Temperature

- 1. Understanding the eBook Teacher Answer Key Explore Learning Gizmo Solubility And Temperature
 - The Rise of Digital Reading Teacher Answer Key Explore Learning Gizmo Solubility And Temperature
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Teacher Answer Key Explore Learning Gizmo Solubility And Temperature
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Teacher Answer Key Explore Learning Gizmo Solubility And Temperature
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Teacher Answer Key Explore Learning Gizmo Solubility And Temperature
 - Personalized Recommendations
 - Teacher Answer Key Explore Learning Gizmo Solubility And Temperature User Reviews and Ratings
 - Teacher Answer Key Explore Learning Gizmo Solubility And Temperature and Bestseller Lists
- 5. Accessing Teacher Answer Key Explore Learning Gizmo Solubility And Temperature Free and Paid eBooks
 - Teacher Answer Key Explore Learning Gizmo Solubility And Temperature Public Domain eBooks
 - Teacher Answer Key Explore Learning Gizmo Solubility And Temperature eBook Subscription Services
 - Teacher Answer Key Explore Learning Gizmo Solubility And Temperature Budget-Friendly Options
- 6. Navigating Teacher Answer Key Explore Learning Gizmo Solubility And Temperature eBook Formats

- o ePub, PDF, MOBI, and More
- Teacher Answer Key Explore Learning Gizmo Solubility And Temperature Compatibility with Devices
- Teacher Answer Key Explore Learning Gizmo Solubility And Temperature Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Teacher Answer Key Explore Learning Gizmo Solubility And Temperature
 - Highlighting and Note-Taking Teacher Answer Key Explore Learning Gizmo Solubility And Temperature
 - Interactive Elements Teacher Answer Key Explore Learning Gizmo Solubility And Temperature
- 8. Staying Engaged with Teacher Answer Key Explore Learning Gizmo Solubility And Temperature
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Teacher Answer Key Explore Learning Gizmo Solubility And Temperature
- 9. Balancing eBooks and Physical Books Teacher Answer Key Explore Learning Gizmo Solubility And Temperature
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Teacher Answer Key Explore Learning Gizmo Solubility And Temperature
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Teacher Answer Key Explore Learning Gizmo Solubility And Temperature
 - Setting Reading Goals Teacher Answer Key Explore Learning Gizmo Solubility And Temperature
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Teacher Answer Key Explore Learning Gizmo Solubility And Temperature
 - Fact-Checking eBook Content of Teacher Answer Key Explore Learning Gizmo Solubility And Temperature
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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unlike a normal chemical reaction that forms molecules nuclear reactions result in the transformation of one element into another this property of nuclear reactions is used in nuclear power plants to gather nuclear energy the three common types of radiations are explained below

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in all accelerators the particles move in a vacuum to avoid collisions with gas molecules when neutrons are required for transmutation reactions they are usually obtained from radioactive decay reactions or from various nuclear reactions occurring in nuclear reactors

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