

TEACHING TRANSPARENCY 22

Microscopic Chapter 2
Section 1.1

Formation of ions

- Q 1. What are cations in the following context?
Electron deficient
- Q 2. How do cations appear in the last table of the last experiment (anion)? Chlorine (anion)
positive electron deficient cation
- Q 3. What is the reason for the loss of electron in cations? Electron deficiency, positive (anion) (anion)
Electron deficiency
- Q 4. What kind of change does the formation of anion cause a shift in the electron? What is the reason for this?
Electron deficient cation
- Q 5. What kind of change does the formation of anion cause a shift in the electron? What is the reason for this?
Electron deficient cation
- Q 6. In the case of cation, the formation of anion is a shift in the electron? What is the reason for this?
Electron deficient cation
- Q 7. In the case of cation, the formation of anion is a shift in the electron? What is the reason for this?
Electron deficient cation
- Q 8. In the case of cation, the formation of anion is a shift in the electron? What is the reason for this?
Electron deficient cation
- Q 9. In the case of cation, the formation of anion is a shift in the electron? What is the reason for this?
Electron deficient cation
- Q 10. In the case of cation, the formation of anion is a shift in the electron? What is the reason for this?
Electron deficient cation
- Q 11. What is the electron deficiency of the following? What is the reason for this?
Electron deficient cation
- Q 12. What is the electron deficiency of the following? What is the reason for this?
Electron deficient cation

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