



## Trends in Atomic Radii

1. Describe and explain the trend in atomic radii as you go down group 2.
2. Describe and explain the trend in atomic radii as you go across period 3.
3. Describe and explain the trend in atomic radii as you go across period 4 from Scandium to Zinc.

## Radii Comparisons

4. Compare the radii of the following pairs, including both a description and an explanation in your answer.
  - a. Na and  $\text{Na}^+$
  - b. Li and  $\text{O}^-$
  - c.  $\text{Al}^{3+}$  and  $\text{Na}^+$
  - d.  $\text{H}^+$  and  $\text{H}^-$
  - e.  $\text{S}^{2-}$  and  $\text{Cl}^-$
  - f. F and  $\text{Na}^+$



Answers

# Teaching Transparency Atomic And Ionic Radii

**IM Harris**



## **Teaching Transparency Atomic And Ionic Radii:**

## Decoding **Teaching Transparency Atomic And Ionic Radii**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Teaching Transparency Atomic And Ionic Radii**," a mesmerizing literary creation penned by a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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