

2. Do you think a chemical reaction has taken place? Explain.

Yes, I do think that the gas discharged into the air caused a chemical reaction to occur.

<b>Activity A:</b>  <b>Observing chemical changes</b>	<b>Get the Gizmo ready:</b> <ul style="list-style-type: none"><li>Click Reset (↻). Check that the reactants are still Sodium and Water.</li><li>Turn on the Label reactants checkbox.</li></ul>	
---	---	---

**Introduction:** It is important to distinguish chemical changes, in which new substances are formed, from **physical changes**, which do not create new substances. In this activity, you will look at many kinds of evidence that chemists use to see if a chemical change has taken place.

**Question:** What kinds of evidence indicate a chemical change has taken place?

1. **Observe:** Some chemical reactions release heat, and others absorb heat. In an **exothermic** reaction, heat is released and the temperature of the system rises. In an **endothermic** reaction, heat is absorbed and the temperature of the system decreases.

In the Gizmo, drag the Thermometer into the flask of water.

- A. What is the starting temperature?
- B. Click Play, and wait for the reaction to end. What is the final temperature?
- C. Was this reaction exothermic or endothermic?

21 Celsius

59 Celsius

Exothermic.

2. **Observe:** Two families of chemicals are **acids** and **bases**. Acids and bases can be detected by an **indicator**, which is a substance that changes color in the presence of an acid or a base. Phenol red is an indicator that is yellow in an acid, orange in a neutral solution, and pink in a base.

- A. Click Reset. Drag the Phenol red next to the flask of water. What does the indicator show?...

Start PH: Neutral, 423.0g mass. Orange Phenol.

- B. Click Play, and wait for the reaction to end. What does the indicator show now?

PH: Basic, 422.9g mass, sodium dissolved. When the water from the sodium was added, phenol turned red or pink.

3. **Observe:** Click Reset. Select the Gas collection setup. Chemists use this apparatus to collect any gases produced in the reaction. From the reaction flask, gases travel through a long tube and into a cylinder of water. As gases bubble into the cylinder, the water is displaced (removed) until the cylinder is filled with gas.

Click Play and observe the cylinder. Was any gas produced in the reaction?

# Study Of Chemical Changes Lab Answer Key

**K Morrison**

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