Cracking the Periodic Table Code

Why aren't the elements listed in alphabetical order?

Why?

As charts go, the periodic table is a bit odd. It's not square. Large portions of the table appear to be missing at the top. It's not organized alphabetically so elements can be found easily. But to a chemist, the periodic table is a very powerful tool. The periodic table is organized by properties, both chemical and physical. Those properties relate to the electronic structure of the atoms of each element. In today's activity, you are going to study how the ground state electron configurations and structure of atoms are related to the shape and organization of the periodic table.

Blank Periodic Table

	_		
	-	3p ³	

1. Circle the number of the team your group has been assigned.

Team 1: H, Li, Na, K, Rb, Cs

Team 2: Fe, Co, Ni, Ru, Rh, Pd

Team 3: Ne, Ar, Kr, Xe, Rn

Team 4: F, Cl, Br, I, At

Team 5: Be, Mg, Ca, Sr, Ba

Team 6: N, P, As, Sb, Bi

Team 7: Ti, V, Cr, Zr, Nb, Mo

Team 8: honors only Er, Tm, Yb, Fm, Md, No

Team 9: honors only Sm, Tb, Ho, Pu, Bk, Es

Teacher Guide Cracking The Periodic Table Code

Amin Akhnoukh, Kamil Kaloush, Mena Souliman, Carlos Chang

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