REVIEW OF BASIC CHEMISTRY ANSWER KEY

Name the following elements. Spelling counts:

Write the symbols for the following elements.

H	hydrogen
S	sulphur
CI	chlorine
C	carbon
N	nitrogen
Na	sodium
Pb	lead
Ag	silver

sodium	PMan
phosphorus	P
fluorine	-
magnesium	Mg
potassium	K
calcium	Car
zina	Zn
iron	Fo

Using a periodic table, record the atomic number for the following elements:

gold (Au) 79 copper (Cu) 29

 Using a periodic table, record the atomic mass for the following elements to one decimal place:

chlorine (CI) 35.5 calcium (Ca) 40.1

 Complete the following table. Use the information provided in the chart – not a periodic table – to determine atomic masses, but you may refer to a periodic table to name the element.

Element	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
Lead	82	207	82	125	82
Argon	18	58	18	40	19
Barium	56	137	56	81	56
Tin	50	112	50	62	50

6. Match the unit with its description:

_F	302	amount of substance	A.	joule
C	2.	density	в.	g
В	3.	mass	G.	g/mL
D	4.	molar mass	D.	g/mok
A_	5.	energy	€.	L
E	6.	volume	F.	mole

Teachersweb Chemistry Answer

Francis X. Sutman, Sidney P. Harris

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