

Water & Aqueous Solutions – Practice Problems

Exercises

1. Calculate the pH of the following aqueous solutions:
 (a) 0.1 M HCl
 (b) 0.01 M HCl
 (c) 0.001 M HCl
 (d) 0.1 M NaOH
 (e) 0.01 M NaOH
 (f) 0.001 M NaOH
2. Calculate the pH of the following aqueous solutions:
 (a) 0.1 M H_2SO_4
 (b) 0.01 M H_2SO_4
 (c) 0.001 M H_2SO_4
 (d) 0.1 M HNO_3
 (e) 0.01 M HNO_3
 (f) 0.001 M HNO_3
3. Calculate the pH of the following aqueous solutions:
 (a) 0.1 M H_2CO_3
 (b) 0.01 M H_2CO_3
 (c) 0.001 M H_2CO_3
 (d) 0.1 M H_2PO_4^-
 (e) 0.01 M H_2PO_4^-
 (f) 0.001 M H_2PO_4^-
4. Calculate the pH of the following aqueous solutions:
 (a) 0.1 M H_2S
 (b) 0.01 M H_2S
 (c) 0.001 M H_2S
 (d) 0.1 M HS^-
 (e) 0.01 M HS^-
 (f) 0.001 M HS^-
5. Calculate the pH of the following aqueous solutions:
 (a) 0.1 M $\text{H}_2\text{C}_2\text{O}_4$
 (b) 0.01 M $\text{H}_2\text{C}_2\text{O}_4$
 (c) 0.001 M $\text{H}_2\text{C}_2\text{O}_4$
 (d) 0.1 M HC_2O_4^-
 (e) 0.01 M HC_2O_4^-
 (f) 0.001 M HC_2O_4^-
6. Calculate the pH of the following aqueous solutions:
 (a) 0.1 M $\text{H}_2\text{C}_2\text{O}_4$
 (b) 0.01 M $\text{H}_2\text{C}_2\text{O}_4$
 (c) 0.001 M $\text{H}_2\text{C}_2\text{O}_4$
 (d) 0.1 M HC_2O_4^-
 (e) 0.01 M HC_2O_4^-
 (f) 0.001 M HC_2O_4^-
7. Calculate the pH of the following aqueous solutions:
 (a) 0.1 M $\text{H}_2\text{C}_2\text{O}_4$
 (b) 0.01 M $\text{H}_2\text{C}_2\text{O}_4$
 (c) 0.001 M $\text{H}_2\text{C}_2\text{O}_4$
 (d) 0.1 M HC_2O_4^-
 (e) 0.01 M HC_2O_4^-
 (f) 0.001 M HC_2O_4^-
8. Calculate the pH of the following aqueous solutions:
 (a) 0.1 M $\text{H}_2\text{C}_2\text{O}_4$
 (b) 0.01 M $\text{H}_2\text{C}_2\text{O}_4$
 (c) 0.001 M $\text{H}_2\text{C}_2\text{O}_4$
 (d) 0.1 M HC_2O_4^-
 (e) 0.01 M HC_2O_4^-
 (f) 0.001 M HC_2O_4^-
9. Calculate the pH of the following aqueous solutions:
 (a) 0.1 M $\text{H}_2\text{C}_2\text{O}_4$
 (b) 0.01 M $\text{H}_2\text{C}_2\text{O}_4$
 (c) 0.001 M $\text{H}_2\text{C}_2\text{O}_4$
 (d) 0.1 M HC_2O_4^-
 (e) 0.01 M HC_2O_4^-
 (f) 0.001 M HC_2O_4^-
10. Calculate the pH of the following aqueous solutions:
 (a) 0.1 M $\text{H}_2\text{C}_2\text{O}_4$
 (b) 0.01 M $\text{H}_2\text{C}_2\text{O}_4$
 (c) 0.001 M $\text{H}_2\text{C}_2\text{O}_4$
 (d) 0.1 M HC_2O_4^-
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 (f) 0.001 M HC_2O_4^-

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