

10.1 Periodic Trends



PERIODIC TRENDS: FROM CONCEPT TO PRACTICE

Section 10.1 continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

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Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 1. It increases in the periodic table.
- _____ 2. It decreases in the periodic table.
- _____ 3. Group 1 elements.
- _____ 4. Elements that are shiny and conduct electricity.
- _____ 5. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word as shown in **Model 10.1**.

- _____ 1. *There are more representative elements of elements.*
- _____ 2. *More than three-fourths of the elements in the periodic table are metals.*
- _____ 3. *Group 1A elements except for hydrogen are metals in the solid state.*
- _____ 4. *Group 2A elements are the alkaline earth metals.*
- _____ 5. *Group 7A elements are highly reactive nonmetals known as halogens.*
- _____ 6. *Group 8A elements are inert nonreactive elements known as noble gases.*
- _____ 7. *Elements from groups of both metals and nonmetals are metalloids.*

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Rachel S Tattersall



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