

## Biology – Chapter 5 Study Guide

### Overview: The Molecules of Life

- Within cells, small organic molecules are joined together to form larger molecules
- These large macromolecules may consist of thousands of covalently bonded atoms and weigh more than 100,000 Daltons.
- The four major classes of macromolecules are carbohydrates, lipids, proteins, and nucleic acids

### Concept 5.1 Most macromolecules are polymers, built from monomers

- Three of the four classes of macromolecules – carbs, proteins, and nucleic acids – form chainlike molecules called polymers
  - A polymer is a long molecule consisting of many similar or identical building blocks linked by covalent bonds
  - Their repeated units are small molecules called **monomers**
  - Some of the molecules that serve as monomers have other functions of their own
- The chemical mechanisms that cells use to make and break polymers are similar for all classes of macromolecules
- Monomers are connected by covalent bonds that form through the loss of a water molecule. This reaction is called a **condensation reaction** or **dehydration reaction**
  - When a bond forms between two monomers, each monomer contributes part of the water molecule that is lost. One monomer provides a hydroxyl group (-OH), while the other provides a hydrogen (-H).
  - Cells invest energy to carry out dehydration reactions
  - The process is aided by enzymes
- The covalent bonds connecting monomers in a polymer are disassembled by **hydrolysis**, a reaction that is effectively the reverse of dehydration
  - In hydrolysis, bonds are broken by the addition of water molecules. A hydrogen atom attaches to one monomer, and a hydroxyl group attaches to the adjacent monomer
  - Our food is taken in as organic polymers that are too large for our cells to absorb. Within the digestive tract, various enzymes direct hydrolysis of specific polymers. The resulting monomers are absorbed by the cells lining the gut and transported to the bloodstream for distribution to body cells
  - The body cells then use dehydration reaction to assemble the monomers into new polymers that carry out function specific to the particular cell type

*An immense variety of polymers can be built from a small set of monomers*

- Each cell has thousands of different kinds of macromolecules
  - These molecules vary among cells of the same individual. They vary more among unrelated individuals of a species, and even more between species
- This diversity comes from various combinations of the 40-50 common monomers and some others that occur rarely
  - These monomers can be connected in a great many combinations, just as the 26 letters in the alphabet can be used to create a great diversity of words

### Concept 5.2 Carbohydrates serve as fuel and building material

- Carbohydrates include sugars and their polymers

# Study Guide Modern Biology Ch 5 Answer

**Martha Mackin**



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**Modern Biology** James Howard Otto,1985      **Life: The Science of Biology Study Guide** William K. Purves,Edward Dzialowski,Lindsay Goodloe,Betty McGuire,Nancy Guild,Paula Mabee,2003-12-26 New edition of a text presenting underlying concepts and showing their relevance to medical agricultural and environmental issues Seven chapters discuss the cell information and heredity evolutionary process the evolution of diversity the biology of flowering plants and of animals and ecology and biogeography Topics are linked by themes such as evolution the experimental foundations of knowledge the flow of energy in the living world the application and influence of molecular techniques and human health considerations Includes a CD ROM which covers some of the subject matter and introduces and illustrates 1 700 plus key terms and concepts Annotation copyrighted by Book News Inc Portland OR      *Biology I* Alfred E. Zietlow,1963      **Study Guide for 31840 - Biology-First Edition** Neil A. Campbell,1987      **Life Study Guide** David E. Sadava,Gordon H. Orians,Craig Heller,William K. Purves,2006-12-22 Especially helpful for AP Biology students each chapter of the study guide offers a variety of study and review tools The contents of each chapter are broken down into both a detailed review of the Important Concepts covered and a boiled down Big Picture snapshot The guide also covers study strategies common problem areas and provides a set of study questions both multiple choice and short answer      *Study Guide for Psychology, Third Edition* Cornelius Rea,Don H. Hockenbury,Sandra E. Hockenbury,2002-08-02 New edition of the Hockenburys text which draws on their extensive teaching and writing experiences to speak directly to students who are new to psychology      **Biology II** Alfred E. Zietlow,1963

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about biological information they encounter in the media The completely revised content and exceptional new art and photos provide students with a more user friendly text while excellent learning tools maximize comprehension of material

*Notebook On Structural Biology Techniques: Study Guide For Practitioners* Ipsita Banerjee, 2025-05-29 This book describes the basic techniques required in learning structural biology Apart from the classical and standard approaches like X ray crystallography and SAXS small angle X ray scattering it has quite an extensive chapter on the most recent approach namely cryo Electron Microscopy The material is partly pedagogical allowing readers to learn the subject including some advanced topics The application of these techniques to a specific example Ribonucleotide Reductase is discussed At the end there is a detailed Mathematical Supplement broken into six sections that develop from basic principles the various mathematical machinery needed to learn the techniques of structural biology It elaborates on the key mathematical concepts mentioned in the various chapters Additionally there is a collection of Ipsita s papers highlighting her seminal contributions especially in the data processing of cryo EM

*Study Guide to Accompany Basics for Chemistry* Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

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