

Now _____ Date _____ Class _____

CHAPTER 9 **STUDY GUIDE FOR CONTENT MASTERY**

Covalent Bonding

Section 9.1 The Covalent Bond

In your textbook, read about the nature of covalent bonds.

Use each of the terms below just once to complete the passage.

covalent bond	molecule	sigma bond	exothermic	p bond
---------------	----------	------------	------------	--------

When sharing of electron occurs, the attachment between atoms that results is called a(n) **(1) covalent bond**. When such an attachment is formed, bond dissociation energy is released, and the process is **(2) exothermic**. When two or more atoms bond by means of electron sharing, the resulting particle is called a(n) **(3) molecule**. If the electrons shared are contained between the two atoms, the attachment is called a(n) **(4) sigma bond**. If the sharing involves the overlap of parallel orbitals, the attachment is called a(n) **(5) p bond**.

In your textbook, read about single and multiple bonds and bond strengths.

Circle the letter of the choice that best completes the statement or answers the question.

- Is what form the elements such as hydrogen, nitrogen, and oxygen normally exist?
 - as single atoms
 - ☒ as molecules containing three atoms
 - as molecules containing two atoms
 - as molecules containing four atoms
- How many electrons are shared in a double covalent bond?
 - none
 - one
 - two
 - four
- Bond length is the distance between
 - two molecules of the same substance
 - ☒ the nuclei of two attached atoms
 - the electrons in two attached atoms
 - the orbitals of two attached atoms
- Which of the following relationships relating to bond length is generally correct?
 - ☒ the shorter the bond, the stronger the bond
 - the shorter the bond, the weaker the bond
 - the shorter the bond, the lower the electron is in
 - the shorter the bond, the lower the bond dissociation energy

Study Guide for Content Mastery

Chemistry: Matter and Change • Chapter 9

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CHAPTER 9 **STUDY GUIDE FOR CONTENT MASTERY**

Section 9.2 Naming Molecules

In your textbook, read about how binary compounds and acids are named from their formulas.

For each statement below, write true or false.

- Binary molecular compounds are generally composed of a metal and a nonmetal. **false**
- The second element in the formula of a binary compound is named using the suffix -ite. **false**
- The prefix zero indicates three atoms. **false**
- The prefix four indicates six atoms. **true**
- In naming the first element in a formula, the prefix zero is not used. **true**
- For binary acids, the hydrogen part of the compound is named using the prefix hydro. **true**
- An oxyacid contains only two elements. **false**
- If the name of the anion of an oxyacid ends in -ite, the acid name contains the suffix -ous. **false**

In your textbook, read about naming molecular compounds and oxoacids.

For each item in Column A, write the letter of the matching item in Column B.

Column A

- C_2H_6
- H_2CO_3
- HNO_3
- H_2SO_4
- H_2SO_3
- $\text{H}_2\text{N}_2\text{O}_2$
- H_2SO_4
- HNO_3
- H_2SO_3
- HNO_2
- H_2SO_4

Column B

- hydrobromic acid
- dinitrogen tetroxide
- carbon monoxide
- nitrous acid
- ammonia
- acetic acid
- carbonic acid
- bromic acid
- carbon dioxide

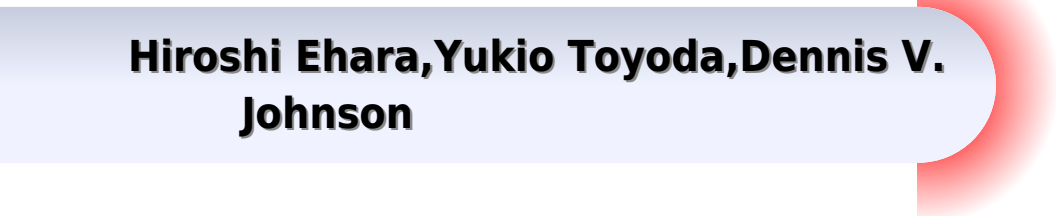
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Study Guide For Content Mastery Molecular Shape

**Hiroshi Ehara, Yukio Toyoda, Dennis V.
Johnson**



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extraction and manufacture characteristics and utilization of starch and cultural anthropological and folkloristic aspects Problems such as food shortages due to increasing populations global warming and climate change and decreasing reserves of oil and other underground resources have become more pressing in recent years In the context of these problems the book examines the role of the sago palm in sustainable food production in the manufacture of other foodstuffs as a raw material for ethanol and in the manufacture of biodegradable plastics In addition to academics this book will be useful to researchers and government officials working for international agencies national governments municipalities and other research organizations technicians researchers managers entrepreneurs and others working in industries such as agriculture plant production food production manufacturing chemical engineering energy production and distribution **Mastering**

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Physiology and Pathophysiology Nancy Tkacs, Nancy C. Tkacs, Linda Herrmann, Randall Johnson, 2020-03-26 Note to Readers Publisher does not guarantee quality or access to any included digital components if book is purchased through a third party seller Specifically designed for future healthcare providers who will diagnose manage and prescribe This advanced physiology and pathophysiology text is designed to address the specific learning needs of future nurse practitioners physician assistants and other advanced healthcare providers caring for patients across the lifespan Focusing on practical applications of physiology it facilitates in depth understanding of important pathophysiological concepts as they relate to major disorders commonly seen in clinical practice and includes comprehensive pediatric and geriatric considerations This knowledge is crucial to providing the foundation required to be an informed and confident clinical decision maker The author team includes experienced clinicians and educators nurses and nurse practitioners physician assistants doctors of pharmacy physicians and basic scientists This collaboration has produced a text that carefully details and richly illustrates the cellular

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