

10.1 Periodic Trends



PERIODIC TRENDS FROM CONCEPTUAL MASTERY

Section 10.1 (continued)

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	19
_____	20
_____	Aluminum
_____	14-18
_____	Transition?

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 19. It exists in the periodic table.
- _____ 20. It has no line periodic table.
- _____ 21. Group 1 elements.
- _____ 22. Elements that are strong oxidizing agents.
- _____ 23. Group 18 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- _____ 24. There are two representative elements of elements.
- _____ 25. Most elements have higher of the elements in the periodic table are metals.
- _____ 26. Group 1A elements except for hydrogen are metals in the solid state.
- _____ 27. Group 2A elements are the alkaline earth metals.
- _____ 28. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 29. Group 8A elements are very reactive elements known as noble gases.
- _____ 30. Alkaloids have properties of both metals and nonmetals.

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1. Understanding the eBook Study Guide For Content Mastery Neutralization
 - The Rise of Digital Reading Study Guide For Content Mastery Neutralization
 - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide For Content Mastery Neutralization
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide For Content Mastery Neutralization
 - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide For Content Mastery Neutralization
 - Personalized Recommendations
 - Study Guide For Content Mastery Neutralization User Reviews and Ratings
 - Study Guide For Content Mastery Neutralization and Bestseller Lists
5. Accessing Study Guide For Content Mastery Neutralization Free and Paid eBooks

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- Study Guide For Content Mastery Neutralization eBook Subscription Services
- Study Guide For Content Mastery Neutralization Budget-Friendly Options
- 6. Navigating Study Guide For Content Mastery Neutralization eBook Formats
 - ePub, PDF, MOBI, and More
 - Study Guide For Content Mastery Neutralization Compatibility with Devices
 - Study Guide For Content Mastery Neutralization Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Study Guide For Content Mastery Neutralization
 - Highlighting and Note-Taking Study Guide For Content Mastery Neutralization
 - Interactive Elements Study Guide For Content Mastery Neutralization
- 8. Staying Engaged with Study Guide For Content Mastery Neutralization
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Study Guide For Content Mastery Neutralization
- 9. Balancing eBooks and Physical Books Study Guide For Content Mastery Neutralization
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Study Guide For Content Mastery Neutralization
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Study Guide For Content Mastery Neutralization
 - Setting Reading Goals Study Guide For Content Mastery Neutralization
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Study Guide For Content Mastery Neutralization
 - Fact-Checking eBook Content of Study Guide For Content Mastery Neutralization
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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