

# CHAPTER 14 STUDY GUIDE

## Mixtures and Solutions

### Section 14.1 Heterogeneous and Homogeneous Mixtures

*At your notebook, read about suspensions and colloids.*

*For each statement below, write true or false.*

- \_\_\_\_\_ 1. A solution is a mixture consisting particles that settle out of the mixture if left undisturbed.
- \_\_\_\_\_ 2. The most abundant substance in a colloid is the dispersion medium.
- \_\_\_\_\_ 3. A colloid can be separated by filtration.
- \_\_\_\_\_ 4. A solid emulsion consists of a liquid dispersed in a solid.
- \_\_\_\_\_ 5. Whipped cream is an example of a foam.
- \_\_\_\_\_ 6. In an aerosol, the dispersing medium is a liquid.
- \_\_\_\_\_ 7. Brownian motion results from the collisions of particles of the dispersion medium with the dispersed particles.
- \_\_\_\_\_ 8. Dispersed particles in a colloid do not tend to settle out because they have polar or charged atomic groups on their surfaces.
- \_\_\_\_\_ 9. Adding an electrolyte into a colloid stabilizes the colloid.
- \_\_\_\_\_ 10. Colloids demonstrate the Tyndall effect.

The table below lists the characteristics of particles in colloids, solutions, and suspensions. Place a check in the column of each mixture where particles have a particular characteristic.

Characteristics of Particles	Colloid	Solution	Suspension
01. Less than 1 nm in diameter			
02. Between 1 nm and 1000 nm in diameter			
03. More than 1000 nm in diameter			
04. Settle out if undisturbed			
05. Pass through standard filter paper			
06. Largest vapor pressure			
07. Scatter light			

# Study Guide For Content Mastery Section 12 4 Percent Yield

**Jianjun Gao**



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