

Introduction to Chemistry

Section 1.1 The Stories of Two Chemicals

In your textbook, read about the ozone layer.

Use each of the boxes below just once to complete the passage.

atmosphere	oxygen gas	ozone	ionized state
ionosphere	hydroxyls	ultraviolet radiation	

Earth's **OZ** _____ is made up of several layers. The air ionosphere makes up the lowest level. This layer is called the **OZ** _____. The next layer up is called the **OZ** _____. This layer contains a protective _____.

UV _____ rays

cause harm when **UV** _____ is caused by ultraviolet radiation in the upper part of the atmosphere. The ozone layer's layer protects Earth, without allowing _____.

UV _____ Without ozone, you get more likely to get a rash or possibly skin cancer. The thinning of the ozone layer called the **OZ** _____ is measured because without ozone all organisms on Earth are subject to harm from too much radiation.

In your textbook, read about chlorofluorocarbons.

Use each statement below, replacing as follows.

- _____ **1.** CFCs is another name for a chlorofluorocarbon.
- _____ **2.** CFCs are made up of carbon, fluorine, and chlorine.
- _____ **3.** All CFCs are synthetic chemicals.
- _____ **4.** CFCs usually react slowly with other chemicals.
- _____ **5.** CFCs were developed as replacements for toxic refrigerants.

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Table of Contents Study Guide For Content Mastery Chapter 3

1. Understanding the eBook Study Guide For Content Mastery Chapter 3
 - The Rise of Digital Reading Study Guide For Content Mastery Chapter 3
 - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide For Content Mastery Chapter 3
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide For Content Mastery Chapter 3
 - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide For Content Mastery Chapter 3

- Personalized Recommendations
 - Study Guide For Content Mastery Chapter 3 User Reviews and Ratings
 - Study Guide For Content Mastery Chapter 3 and Bestseller Lists
5. Accessing Study Guide For Content Mastery Chapter 3 Free and Paid eBooks
- Study Guide For Content Mastery Chapter 3 Public Domain eBooks
 - Study Guide For Content Mastery Chapter 3 eBook Subscription Services
 - Study Guide For Content Mastery Chapter 3 Budget-Friendly Options
6. Navigating Study Guide For Content Mastery Chapter 3 eBook Formats
- ePub, PDF, MOBI, and More
 - Study Guide For Content Mastery Chapter 3 Compatibility with Devices
 - Study Guide For Content Mastery Chapter 3 Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Study Guide For Content Mastery Chapter 3
 - Highlighting and Note-Taking Study Guide For Content Mastery Chapter 3
 - Interactive Elements Study Guide For Content Mastery Chapter 3
8. Staying Engaged with Study Guide For Content Mastery Chapter 3
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Study Guide For Content Mastery Chapter 3
9. Balancing eBooks and Physical Books Study Guide For Content Mastery Chapter 3
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Study Guide For Content Mastery Chapter 3
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Study Guide For Content Mastery Chapter 3
- Setting Reading Goals Study Guide For Content Mastery Chapter 3
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Study Guide For Content Mastery Chapter 3

- Fact-Checking eBook Content of Study Guide For Content Mastery Chapter 3
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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