

10.1 Periodic Trends



PERIODIC TRENDS: FROM CONCEPT TO PRACTICE

Section 10.1 continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	19
_____	20
_____	Aluminum
_____	14-18
_____	Transition?

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 19. It exists in the periodic table.
- _____ 20. It has no line periodic table.
- _____ 21. Group 1 elements.
- _____ 22. Elements that are shiny and conduct electricity.
- _____ 23. Group 18 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word as shown in **Model 10.1**.

- _____ 24. There are two representative elements of aluminum.
- _____ 25. About three times heavier of the elements in the periodic table are transition.
- _____ 26. Group 1A elements except for hydrogen are known as the alkali metals.
- _____ 27. Group 2A elements are the alkaline earth metals.
- _____ 28. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 29. Group 8A elements are very unreactive elements known as noble gases.
- _____ 30. Alkaloids have properties of both metals and more reactive metals.

Study Guide For Content Mastery Stoichiometric Answers

Rachel Sandford



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Programmed Topics in General Chemistry Armine D. Paul, 1971 *Bulletin of the American Ceramic Society*, 1960

The Engineer, 1957-10 Metals Abstracts, 1996 Ceramic Abstracts, 1960

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