

Final Exam Study Guide (General Chemistry I)

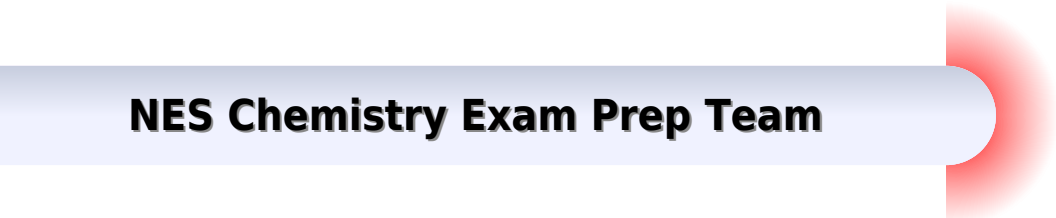
- Chemistry - the study of matter and the changes that matter undergoes
- Matter - anything that has mass and takes up space
 - Three states of matter: solid, liquid, and gas (plasma is a gas)

	Proximity of Particles	Energetics	Size/Shape
Solid	Closely packed and in direct contact with each other	Have very little energy of motion (slow)	Fixed shape and fixed volume
Liquid	In indirect contact with each other; usually about a particles distance apart (bounce off each other)	Have an intermediate amount energy of motion (medium)	Does not have a fixed shape; has a fixed volume
Gas	Particles are very far apart	Have a large amount of energy of motion (fast)	No fixed shape or volume

- **PARTICLES ARE ALWAYS MOVING!**
- "Can you undo it?" (Page 12)
 - Physical Changes - changes physical appearance but not its composition
 - Ex. Melting, evaporation, freezing, etc.
 - Chemical Changes - changes in chemical composition of matter (chemical reaction)
 - Ex. Baking soda and vinegar, burning, mixing, etc.
 - Physical Properties - characteristics that can be described by observing the matter; no change in composition is required to detect a physical property
 - Ex. Color, odor, density, etc.
 - Chemical Properties - a change in composition is required to detect a chemical property
 - Ex. Flammability
- Accuracy Versus Precision:
 - In every measurement, there is some degree of uncertainty.
 - **NUMBERS WITHOUT UNITS ARE MEANINGLESS IN MEASUREMENT!**
 - Accuracy - the agreement of a measurement with a known or accepted value (correctness)
 - Precision - the agreement of a measurement with other measurements in a given set of measurements (reproducibility/repeatability)
 - You can have accuracy without precision, and precision without accuracy.
 - **WE ARE ONLY AS CERTAIN AS OUR LEAST CERTAIN NUMBER!**
 - 254 can mean 253.5 to 254.4
 - 254.0 can mean 253.95 to 254.04
 - 254.00 can mean 253.950 to 254.004
 - 254.000 can mean 253.9500 to 254.0004
- Density - the mathematical relationship of mass to volume

Study Guide For National Chemistry Exam

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Chemistry National Learning Corporation, 2020-03-15 SAT Subject Tests developed by The College Board are required by many colleges and universities as part of their admission process The Chemistry Passbook R includes hundreds of questions similar to those found on the real exam

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CHEMISTRY National Learning Corporation, 2019

Review Guide for LPN/LVN Pre-entrance Exam National League for Nursing, Natl League Nursing, 2008-09-12 Providing an overview of the math science and reading comprehension skills necessary for admission to LPN programs this bestselling guide includes multiple choice questions and practice examinations in each of the three test areas

Organic Chemistry National Learning Corporation, 2020-03-15

National Counselor Exam (NCE) and Counselor Preparation Comprehensive Exam (CPCE) KaRae' NMK Powers-Carey, LoriAnn Sykes Stretch, 2023-03-03 Designed to bolster CPCE and NCE exam success on the first try this unique study guide takes the mystery out of exam preparation by providing concrete strategies for mastering essential information end of chapter quizzes providing prompt reinforcement of content two full length exams mirroring the NCE and CPCE in format and breadth and proven tactics for mitigating test anxiety The resource is organized around the latest exam outline from the NBCC so that

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Clinical Chemistry National Learning Corporation,2020-03-15

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Table of Contents Study Guide For National Chemistry Exam

1. Understanding the eBook Study Guide For National Chemistry Exam
 - The Rise of Digital Reading Study Guide For National Chemistry Exam
 - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide For National Chemistry Exam
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide For National Chemistry Exam
 - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide For National Chemistry Exam
 - Personalized Recommendations
 - Study Guide For National Chemistry Exam User Reviews and Ratings
 - Study Guide For National Chemistry Exam and Bestseller Lists

5. Accessing Study Guide For National Chemistry Exam Free and Paid eBooks
 - Study Guide For National Chemistry Exam Public Domain eBooks
 - Study Guide For National Chemistry Exam eBook Subscription Services
 - Study Guide For National Chemistry Exam Budget-Friendly Options
6. Navigating Study Guide For National Chemistry Exam eBook Formats
 - ePub, PDF, MOBI, and More
 - Study Guide For National Chemistry Exam Compatibility with Devices
 - Study Guide For National Chemistry Exam Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Study Guide For National Chemistry Exam
 - Highlighting and Note-Taking Study Guide For National Chemistry Exam
 - Interactive Elements Study Guide For National Chemistry Exam
8. Staying Engaged with Study Guide For National Chemistry Exam
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Study Guide For National Chemistry Exam
9. Balancing eBooks and Physical Books Study Guide For National Chemistry Exam
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Study Guide For National Chemistry Exam
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Study Guide For National Chemistry Exam
 - Setting Reading Goals Study Guide For National Chemistry Exam
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Study Guide For National Chemistry Exam
 - Fact-Checking eBook Content of Study Guide For National Chemistry Exam
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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