

Organic Chemistry 1: Practice Exam 1

1. What is the ground state electron configuration for each of the elements below?

a) Chlorine (Cl) \_\_\_\_\_ b) Nitrogen (N) \_\_\_\_\_

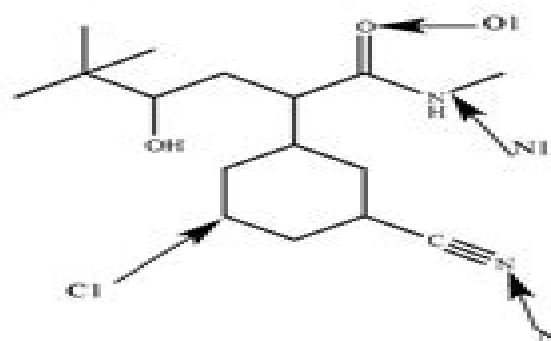
2. Draw the Lewis Structure for each molecule below and show all valence electrons and formal charges.

a)  $(\text{CH}_3)_2\text{CHCHO}$

b)  $\text{N}_2\text{H}_4$

c)  $\text{C}_2\text{H}_2$

3. Examine this suave looking molecule



Cl: Hybridization \_\_\_\_\_  
Bond angles \_\_\_\_\_

O1: Hybridization \_\_\_\_\_  
Bond angles \_\_\_\_\_

N1: Hybridization \_\_\_\_\_  
Bond angles \_\_\_\_\_

N2: Hybridization \_\_\_\_\_  
Bond angles \_\_\_\_\_

# Study Guide For Une Organic Chemistry Exam

**Didier Musso**



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Chemistry Rob Ritchie, 2008-07 Divided into seven manageable day sections this timed revision programme covers essential GCSE topics in double page spreads These spreads indicate how much time should be spent on each section and combine clear and concise explanations flow charts spidergrams and illustrations with progress check questions and answers

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