#### 1.3 Thermodynamics: Study Guide

Use this pulses to review the New vacabulary for this unit.

Match the vocabulary words on the left with the appropriate definitions and units on the right.

a Thermalthorps	A kinetic energy in transit from one oldestice
	another due to temperature difference
	L. Degrees
is Temperature	a. louies
	Ja., The over age Minetic energy of particles in
	an abused

2. In the habite below, write in the anim sociate becoming

Rate of Energy Transfer	Thermal Construction	Coefficient of
P. EAST or P. C.	K- PL	Conductivity
	P. EAST or P. C.	P. EAST or P. C. Conductivity

- For each example, write the correct number for the law of thermodynamics it describes (0, 1, 2).
  - the law of everyone
  - b. \_\_\_\_\_ Thermal energy can only be increased in a system by adding heat from an external source or performing work within the system.
  - If two systems are separately found to be in thermal equilibrium with arthird, then the two systems are in thermal equilibrium with each other.
  - d \_\_\_\_\_ Thermal energy flows only from warmer to cooler
  - The law of conservation of energy applied to thermal ty items
- How much thermal energy would it take to naise the temperature of 11 (193 of water by 20°C? The specific heat capacity of water is 4164 and 72°

=149 -4464 2 - 10'9



0- 41 440 1

A materials scientist is trying to determine the specific heat dispatity of an experimental material.
She applies 30 id of charmal energy to a Sing sample of the material and records a 24°C temperature increase in the sample. What is the specific heat capacity of the material?





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**Sundar Nathan** 

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