

Chapter 39 The Atomic Nucleus and Radioactivity

Exercises

39.1 The Atomic Nucleus (pages 783–784)

- Circle the letter of each nucleon.
☒ a. proton ☒ b. neutron
c. electron d. positron
- Is the following sentence true or false? The mass of a proton and an electron are nearly the same. false
- Most of an atom's mass is located in its nucleus.
- Describe the relationship between the number of protons and the number of electrons in a neutral atom.
They are equal.
- Circle the letter of the particle that determines the kind of element and the chemical properties of an atom.
☒ a. proton b. neutron
c. electron d. positron
- Describe the primary role of the neutrons in the nucleus of an atom.
They act as glue, holding the nucleus together.
- Circle the letter that best describes the strong force.
a. an attractive magnetic force
b. a repulsive electric force
☒ c. an attractive nuclear force
d. an attractive gravitational force
- Is the following sentence true or false? The strong force acts across large distances. false
- The two major forces that act on a nucleus are the strong force and electrical forces.
- A nucleus needs a certain ratio of protons to neutrons to make it stable.

39.2 Radioactive Decay (pages 785–786)

- What will a lone neutron decay into?
a proton and an electron (plus an antineutrino)
- Particles that decay spontaneously and emit particles and energy are said to be radioactive.
- Explain what is meant by saying that radioactivity is governed by mass-energy equivalence.
Particles decay only when their combined products have less mass after decay than before.
- All atoms with atomic numbers greater than 83 decay in one way or another.

The Nucleus Chapter 30 Study Guide Answer

Joseph J. Boyle, Douglas C. Giancoli



The Nucleus Chapter 30 Study Guide Answer:

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