

Unit 2 Worksheet 3 – P&Tn Problems

On each of the problems below, show what the given P, V, T, or n of State 1 leads to for State 2 using a strategy. In P, V, T, or n, use an arrow that clearly identifies what that variable is, the operation that it is, and the unit. Create partial diagrams of the initial and final conditions.

1. What would be the new pressure if 1.50 mol of gas at standard pressure is compressed to a volume of 20.0 cm³? (P, V, and T is constant)

	P	V	T	n
Initial State		10.0 cm ³	273 K	
Final State		20.0 cm ³		
n				

2. The pressure of a sample of gas is 1.00 atm at 20 °C in a flask. What is the new volume if the flask is heated, where the temperature is 30 °C. Assume the flask pressure is the same?

	P	V	T	n
Initial State	1.00 atm	10.0 cm ³		
Final State		20.0 cm ³		
n				

3. A sample of gas occupies 1.00 mol at 20 °C. What is the volume when the temperature is increased to 30 °C? (P and n is constant)

	P	V	T	n
Initial State	1.00 mol		273 K	
Final State			273 K	
n				

Unit 2 3 Pvtn Problems Answers

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