

Chemistry – Unit 4 Worksheet 4

More Compound Analysis

Use the following information about the masses of elements in each pair of compounds to help you suggest formulae that account for these ratios. Be sure to show all mathematical work and reasoning and use complete sentences in explanations.

1. Compounds of carbon and oxygen. 100 g each of two compounds of carbon and oxygen were analyzed and found to contain the following:

Compound A: 57.1 g O / 42.9 g C

Compound B: 72.7 g O and 27.3 C

- Determine the value of the ratio $\frac{\text{mass O}}{\text{mass C}}$ in each compound. A ____ B ____
- How does the mass ratio for compound B compare to that in compound A?
- Express these ratios as improper fractions.
- For each hypothesis, sketch particle diagrams for the compounds of A and B that account for these mass ratios. Write the formula for the compound in each diagram.

Hypothesis 1 Atoms of C and O have the same mass	Hypothesis 2 Atoms of O are heavier than C atoms by the ratio in compound A.
A	A
B	B

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