

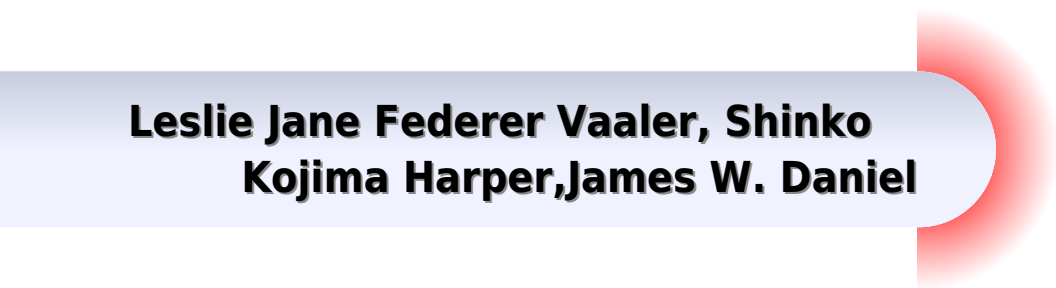
Balancing Equations Worksheet

- 1) $\underline{\hspace{1cm}} \text{H}_3\text{PO}_4 + \underline{\hspace{1cm}} \text{KOH} \rightarrow \underline{\hspace{1cm}} \text{K}_3\text{PO}_4 + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 2) $\underline{\hspace{1cm}} \text{K} + \underline{\hspace{1cm}} \text{B}_2\text{O}_3 \rightarrow \underline{\hspace{1cm}} \text{K}_2\text{O} + \underline{\hspace{1cm}} \text{B}$
- 3) $\underline{\hspace{1cm}} \text{HCl} + \underline{\hspace{1cm}} \text{NaOH} \rightarrow \underline{\hspace{1cm}} \text{NaCl} + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 4) $\underline{\hspace{1cm}} \text{Na} + \underline{\hspace{1cm}} \text{NaNO}_3 \rightarrow \underline{\hspace{1cm}} \text{Na}_2\text{O} + \underline{\hspace{1cm}} \text{N}_2$
- 5) $\underline{\hspace{1cm}} \text{C} + \underline{\hspace{1cm}} \text{S}_8 \rightarrow \underline{\hspace{1cm}} \text{CS}_2$
- 6) $\underline{\hspace{1cm}} \text{Na} + \underline{\hspace{1cm}} \text{O}_2 \rightarrow \underline{\hspace{1cm}} \text{Na}_2\text{O}$
- 7) $\underline{\hspace{1cm}} \text{N}_2 + \underline{\hspace{1cm}} \text{O}_2 \rightarrow \underline{\hspace{1cm}} \text{N}_2\text{O}_5$
- 8) $\underline{\hspace{1cm}} \text{H}_3\text{PO}_4 + \underline{\hspace{1cm}} \text{Mg}(\text{OH})_2 \rightarrow \underline{\hspace{1cm}} \text{Mg}_3(\text{PO}_4)_2 + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 9) $\underline{\hspace{1cm}} \text{NaOH} + \underline{\hspace{1cm}} \text{H}_2\text{CO}_3 \rightarrow \underline{\hspace{1cm}} \text{Na}_2\text{CO}_3 + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 10) $\underline{\hspace{1cm}} \text{KOH} + \underline{\hspace{1cm}} \text{HBr} \rightarrow \underline{\hspace{1cm}} \text{KBr} + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 11) $\underline{\hspace{1cm}} \text{Na} + \underline{\hspace{1cm}} \text{O}_2 \rightarrow \underline{\hspace{1cm}} \text{Na}_2\text{O}$
- 12) $\underline{\hspace{1cm}} \text{Al}(\text{OH})_3 + \underline{\hspace{1cm}} \text{H}_2\text{CO}_3 \rightarrow \underline{\hspace{1cm}} \text{Al}_2(\text{CO}_3)_3 + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 13) $\underline{\hspace{1cm}} \text{Al} + \underline{\hspace{1cm}} \text{S}_8 \rightarrow \underline{\hspace{1cm}} \text{Al}_2\text{S}_3$
- 14) $\underline{\hspace{1cm}} \text{Cs} + \underline{\hspace{1cm}} \text{N}_2 \rightarrow \underline{\hspace{1cm}} \text{Cs}_3\text{N}$
- 15) $\underline{\hspace{1cm}} \text{Mg} + \underline{\hspace{1cm}} \text{Cl}_2 \rightarrow \underline{\hspace{1cm}} \text{MgCl}_2$
- 16) $\underline{\hspace{1cm}} \text{Rb} + \underline{\hspace{1cm}} \text{RbNO}_3 \rightarrow \underline{\hspace{1cm}} \text{Rb}_2\text{O} + \underline{\hspace{1cm}} \text{N}_2$
- 17) $\underline{\hspace{1cm}} \text{C}_6\text{H}_6 + \underline{\hspace{1cm}} \text{O}_2 \rightarrow \underline{\hspace{1cm}} \text{CO}_2 + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 18) $\underline{\hspace{1cm}} \text{N}_2 + \underline{\hspace{1cm}} \text{H}_2 \rightarrow \underline{\hspace{1cm}} \text{NH}_3$
- 19) $\underline{\hspace{1cm}} \text{C}_{10}\text{H}_{22} + \underline{\hspace{1cm}} \text{O}_2 \rightarrow \underline{\hspace{1cm}} \text{CO}_2 + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 20) $\underline{\hspace{1cm}} \text{Al}(\text{OH})_3 + \underline{\hspace{1cm}} \text{HBr} \rightarrow \underline{\hspace{1cm}} \text{AlBr}_3 + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 21) $\underline{\hspace{1cm}} \text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_3 + \underline{\hspace{1cm}} \text{O}_2 \rightarrow \underline{\hspace{1cm}} \text{CO}_2 + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 22) $\underline{\hspace{1cm}} \text{C}_7\text{H}_8 + \underline{\hspace{1cm}} \text{O}_2 \rightarrow \underline{\hspace{1cm}} \text{CO}_2 + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 23) $\underline{\hspace{1cm}} \text{Li} + \underline{\hspace{1cm}} \text{AlCl}_3 \rightarrow \underline{\hspace{1cm}} \text{LiCl} + \underline{\hspace{1cm}} \text{Al}$
- 24) $\underline{\hspace{1cm}} \text{C}_2\text{H}_6 + \underline{\hspace{1cm}} \text{O}_2 \rightarrow \underline{\hspace{1cm}} \text{CO}_2 + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 25) $\underline{\hspace{1cm}} \text{NH}_4\text{OH} + \underline{\hspace{1cm}} \text{H}_3\text{PO}_4 \rightarrow \underline{\hspace{1cm}} (\text{NH}_4)_3\text{PO}_4 + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 26) $\underline{\hspace{1cm}} \text{Rb} + \underline{\hspace{1cm}} \text{P} \rightarrow \underline{\hspace{1cm}} \text{Rb}_3\text{P}$
- 27) $\underline{\hspace{1cm}} \text{CH}_4 + \underline{\hspace{1cm}} \text{O}_2 \rightarrow \underline{\hspace{1cm}} \text{CO}_2 + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 28) $\underline{\hspace{1cm}} \text{Al}(\text{OH})_3 + \underline{\hspace{1cm}} \text{H}_2\text{SO}_4 \rightarrow \underline{\hspace{1cm}} \text{Al}_2(\text{SO}_4)_3 + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 29) $\underline{\hspace{1cm}} \text{Na} + \underline{\hspace{1cm}} \text{Cl}_2 \rightarrow \underline{\hspace{1cm}} \text{NaCl}$
- 30) $\underline{\hspace{1cm}} \text{Rb} + \underline{\hspace{1cm}} \text{S}_8 \rightarrow \underline{\hspace{1cm}} \text{Rb}_2\text{S}$
- 31) $\underline{\hspace{1cm}} \text{H}_3\text{PO}_4 + \underline{\hspace{1cm}} \text{Ca}(\text{OH})_2 \rightarrow \underline{\hspace{1cm}} \text{Ca}_3(\text{PO}_4)_2 + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 32) $\underline{\hspace{1cm}} \text{NH}_3 + \underline{\hspace{1cm}} \text{HCl} \rightarrow \underline{\hspace{1cm}} \text{NH}_4\text{Cl}$
- 33) $\underline{\hspace{1cm}} \text{Li} + \underline{\hspace{1cm}} \text{H}_2\text{O} \rightarrow \underline{\hspace{1cm}} \text{LiOH} + \underline{\hspace{1cm}} \text{H}_2$
- 34) $\underline{\hspace{1cm}} \text{Ca}_3(\text{PO}_4)_2 + \underline{\hspace{1cm}} \text{SiO}_2 + \underline{\hspace{1cm}} \text{C} \rightarrow \underline{\hspace{1cm}} \text{CaSiO}_3 + \underline{\hspace{1cm}} \text{CO} + \underline{\hspace{1cm}} \text{P}$
- 35) $\underline{\hspace{1cm}} \text{NH}_3 + \underline{\hspace{1cm}} \text{O}_2 \rightarrow \underline{\hspace{1cm}} \text{N}_2 + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 36) $\underline{\hspace{1cm}} \text{FeS}_2 + \underline{\hspace{1cm}} \text{O}_2 \rightarrow \underline{\hspace{1cm}} \text{Fe}_2\text{O}_3 + \underline{\hspace{1cm}} \text{SO}_2$
- 37) $\underline{\hspace{1cm}} \text{C} + \underline{\hspace{1cm}} \text{SO}_2 \rightarrow \underline{\hspace{1cm}} \text{CS}_2 + \underline{\hspace{1cm}} \text{CO}$

Unit 7 Balancing Chemical Reactions Worksheet 2

Answer

**Leslie Jane Federer Vaaler, Shinko
Kojima Harper, James W. Daniel**



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Solutions Teacher Planning Pack Core Book 7 David Baker, 2005 This is a major new series developed to provide complete coverage of the framework for teaching mathematics and Medium Term Plan in a highly accessible and modern format *Merrill Chemistry* Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **Solutions Teacher Planning Pack Support Book 7** David Baker, 2005 The only AQA GCSE maths series to be exclusively endorsed and approved by AQA AQA Mathematics for GCSE blends print and electronic resources to provide you with complete reassurance that you have everything you need to deliver the revised 2006 GCSE Mathematics specification *Chemistry* Carson-Dellosa Publishing, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards **Chemistry**, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards **Te HS&T J** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004-02 Jacaranda Nature of Biology 2 VCE Units 3 and 4, LearnON and Print Judith Kinnear, Marjory Martin, Lucy Cassar, Elise Meehan, Ritu Tyagi, 2021-10-29 Jacaranda Nature of Biology Victoria's most trusted VCE Biology online and print resource The Jacaranda Nature of Biology series has been rewritten for the VCE Biology Study Design 2022 2026 and offers a complete and balanced learning experience that prepares students for success in their assessments by building deep understanding in both Key Knowledge and Key Science Skills Prepare students for all forms of assessment Preparing students for both the SACs and exam with access to 1000s of past VCAA exam questions now in print and learnON new teacher only and practice SACs for every Area of Study and much more Videos by experienced teachers Students can hear another voice and perspective with 100s of new videos where expert VCE Biology teachers unpack concepts VCAA exam questions and sample problems For

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evolution and new engineers have a variety of tools at their disposal to tackle their everyday problems Introduction to Software for Chemical Engineers Second Edition provides a quick guide to the use of various computer packages for chemical engineering applications It covers a range of software applications from Excel and general mathematical packages such as MATLAB and MathCAD to process simulators CHEMCAD and ASPEN equation based modeling languages gProms optimization software such as GAMS and AIMS and specialized software like CFD or DEM codes The different packages are introduced and applied to solve typical problems in fluid mechanics heat and mass transfer mass and energy balances unit operations reactor engineering process and equipment design and control This new edition offers a wider view of packages including open source software such as R Python and Julia It also includes complete examples in ASPEN Plus adds ANSYS Fluent to CFD codes Lingo to the optimization packages and discusses Engineering Equation Solver It offers a global idea of the capabilities of the software used in the chemical engineering field and provides examples for solving real world problems Written by leading experts this book is a must have reference for chemical engineers looking to grow in their careers through the use of new and improving computer software Its user friendly approach to simulation and optimization as well as its example based presentation of the software makes it a perfect teaching tool for both undergraduate and master levels

Resources in Education, 1977-05 Petroleum Engineering: Principles, Calculations, and Workflows Moshood Sanni, 2018-09-21 Ein ausführlicher Praxisleitfaden zu Methoden für die Lösung komplexer Probleme in der Erdöltechnik In der Erdöltechnik dominieren bergreifende wissenschaftliche und mathematische Prinzipien Allerdings gibt es immer wieder Lücken zwischen Theorie und praktischer Anwendung Petroleum Engineering Principles Calculations and Workflows stellt Methoden für die Lösung einer Vielzahl praktischer Probleme in der Erdöltechnik vor Jedes Kapitel beschäftigt sich mit einer spezifischen Problemstellung beschreibt Formeln zur Erläuterung der primären Prinzipien dieses Problems und zeigt im Anschluss einfach nachvollziehbare Handreichungen für die praktische Anwendung Hauptmerkmale dieses Bandes Fundierter und integrierter Ansatz für die Lösung inverser Probleme Ausführliche Untersuchung der Abläufe einschließlich Modell und Parametervalidierung Einfache Ansätze für die Lösung komplexer mathematischer Probleme Komplexe Berechnungen die sich mit einfachen Methoden leicht implementieren lassen Überblick über wichtige Herangehensweisen die für die Software und Anwendungsentwicklung notwendig sind Formel und Modellhandreichungen für die Diagnose erstmalige Parametermodellierung Simulation und Regression Petroleum Engineering Principles Calculations and Workflows ist ein wertvolles Referenzwerk für die Praxis und richtet sich an eine breite Zielgruppe Geowissenschaftler Explorationsgeologen und Ingenieure Dieser zugängliche Leitfaden ein fundiertes Nachschlagewerk für die Lösung alltäglicher Probleme in der Erdöltechnik eignet sich ebenfalls gut für Studenten im Hauptstudium Postgraduierte Berater Softwareentwickler und Berufspraktiker *Statistical Mechanics of Phases and Phase Transitions* Steven A. Kivelson, Jack Mingde Jiang, Jeffrey Chang, 2024-04-09 An engaging undergraduate introduction to the statistical mechanics of phase transitions Statistical

mechanics deploys a powerful set of mathematical approaches for studying the thermodynamic properties of complex physical systems This textbook introduces students to the statistical mechanics of systems undergoing changes of state focusing on the basic principles for classifying distinct thermodynamic phases and the critical phenomena associated with transitions between them Uniquely designed to promote active learning Statistical Mechanics of Phases and Phase Transitions presents some of the most beautiful and profound concepts in physics enabling students to obtain an essential understanding of a computationally challenging subject without getting lost in the details Provides a self contained conceptually deep introduction to the statistical mechanics of phases and phase transitions from a modern perspective Carefully leads students from spontaneously broken symmetries to the universality of phase transitions and the renormalization group Encourages student centric active learning suitable for both the classroom and self study Features a wealth of guided worksheets with full solutions throughout the book that help students learn by doing Includes informative appendixes that cover key mathematical concepts and methods Ideal for undergraduate physics majors and beginning graduate students Solutions manual for all end of chapter problems available only to instructors Chemical Engineering, 1958 Te HS&T 2007 Shrt Crs M Holt Rinehart & Winston, 2007 *Mathematical Interest Theory: Third Edition* Leslie Jane Federer Vaaler, Shinko Kojima Harper, James W. Daniel, 2021-04-15 Mathematical Interest Theory provides an introduction to how investments grow over time This is done in a mathematically precise manner The emphasis is on practical applications that give the reader a concrete understanding of why the various relationships should be true Among the modern financial topics introduced are arbitrage options futures and swaps Mathematical Interest Theory is written for anyone who has a strong high school algebra background and is interested in being an informed borrower or investor The book is suitable for a mid level or upper level undergraduate course or a beginning graduate course The content of the book along with an understanding of probability will provide a solid foundation for readers embarking on actuarial careers The text has been suggested by the Society of Actuaries for people preparing for the Financial Mathematics exam To that end Mathematical Interest Theory includes more than 260 carefully worked examples There are over 475 problems and numerical answers are included in an appendix A companion student solution manual has detailed solutions to the odd numbered problems Most of the examples involve computation and detailed instruction is provided on how to use the Texas Instruments BA II Plus and BA II Plus Professional calculators to efficiently solve the problems This Third Edition updates the previous edition to cover the material in the SOA study notes FM 24 17 FM 25 17 and FM 26 17 Environmental Impacts of Waste Paper Recycling Yrjo Virtanen, Sten Nilsson, 2013-07-04 Public concern for the conservation of natural resources and a general awareness of the environmental consequences of waste disposal is reflected in current legislation aimed at reducing waste Recycling is commonly cited as one of the preferred methods of waste reduction and this book summarizes a recent study of paper recycling in Europe which investigated the entire production and disposal process using a life cycle

methodology The results of the study underline the economic and environmental advantages of paper recycling but more controversially they also show how under certain conditions the renewable character and the high energy content of paper seem to make energy recovery more attractive than recycling

Science Interactions Robert W. Avakian, 1995-07-17

Holt Science and Technology Holt Rinehart & Winston, 2004-02

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