

Balancing Equations Worksheet #2

Note – tough ones will have a coefficient above 3, all other problems use coefficients of 3 or fewer.

- 1) ____ Li_3PO_4 + ____ NaOH $\rightarrow$ ____ LiOH + ____ Na_3PO_4
- 2) ____ MgF_2 + ____ Li_2CO_3 $\rightarrow$ ____ MgCO_3 + ____ LiF
- 3) ____ P_4 + ____ O_2 $\rightarrow$ ____ P_2O_3
- 4) ____ RbNO_3 + ____ MgF_2 $\rightarrow$ ____ $\text{Mg}(\text{NO}_3)_2$ + ____ RbF
- 5) ____ AgNO_3 + ____ Cu $\rightarrow$ ____ $\text{Cu}(\text{NO}_3)_2$ + ____ Ag
- 6) ____ CF_4 + ____ Br_2 $\rightarrow$ ____ CBr_4 + ____ F_2
- 7) ____ HCN + ____ CuSO_4 $\rightarrow$ ____ H_2SO_4 + ____ $\text{Cu}(\text{CN})_2$
- 8) ____ GaF_3 + ____ Cs $\rightarrow$ ____ CsF + ____ Ga
- 9) ____ SrS + ____ PtF_2 $\rightarrow$ ____ SrF_2 + ____ PtS
- 10) ____ N_2 + ____ H_2 $\rightarrow$ ____ NH_3
- 11) ____ LiF + ____ Br_2 $\rightarrow$ ____ LiBr + ____ F_2
- 12) ____ $\text{Pb}(\text{OH})_2$ + ____ HCl $\rightarrow$ ____ H_2O + ____ PbCl_2
- 13) ____ GaBr_3 + ____ Na_2CO_3 $\rightarrow$ ____ NaBr + ____ $\text{Ga}_2(\text{CO}_3)_3$ (a tough one)
- 14) ____ CH_4 + ____ O_2 $\rightarrow$ ____ CO_2 + ____ H_2O
- 15) ____ Li_3PO_4 + ____ CaCl_2 $\rightarrow$ ____ LiCl + ____ $\text{Ca}_3(\text{PO}_4)_2$ (also tough)
- 16) ____ Na + ____ Cl_2 $\rightarrow$ ____ NaCl
- 17) ____ Ga + ____ HCl $\rightarrow$ ____ H_2 + ____ GaCl_3 (last tough one)
- 18) ____ N_2 + ____ F_2 $\rightarrow$ ____ NF_3
- 19) ____ SO_2 + ____ Li_2Se $\rightarrow$ ____ SSe_2 + ____ Li_2O
- 20) ____ NH_3 + ____ H_2SO_4 $\rightarrow$ ____ $(\text{NH}_4)_2\text{SO}_4$

Unit 7 Balancing Chemical Reactions Worksheet 2

Answers

**Holt Rinehart & Winston, Holt,
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