#### Skill Practice 19



1. Write the symbol and charges for the following ions:

Example: calcium v Ca

- A/I pfluospfleide m .....
- B) magnesises w \_\_\_\_\_ C) rehidises w

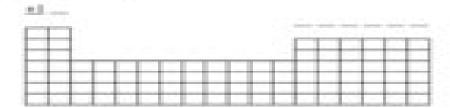
- D) fluoride = E) aluminum = F) sulfide =
- 2. Each of the following formulas is written incorrectly. Please rewrite them correctly. Hyamphr: Ca<sub>2</sub>Cl = CaCl<sub>2</sub>

  - A) BaS = \_\_\_\_\_\_ B) RbN = \_\_\_\_\_ C) LbC = \_\_\_\_\_
  - D) Al<sub>2</sub>N<sub>1</sub> = \_\_\_\_\_ F) Mg/Br<sub>2</sub> = \_\_\_\_ F) O<sub>2</sub>Al<sub>2</sub> = \_\_\_\_

- 3. Write the formulas for the compound formed by the combination of the following pairs of atoms. Example: nitrogen and magnesium a Ma-N-

A) redison and suffer = \_\_\_\_\_\_ B) alaminum and obtains =

- C) phosphorus and calcium = D) burium and oxygen =
- For each of the formulas you wrote in question 3 above, give the names for the compounds. Example: Mg-N<sub>c</sub> w magnesium mbride
- C) D)
- 5. In the blanks above each column, write the charge of the ions formed by atoms in that column. Note: group IA is done for you.



# **Skill Practice 19 Answers Ionic Bonding Practice**

Jay Wackerly, Sarah Zingales, Michael Wentzel, Gautam Bhattacharyya, Brett McCollum

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