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Yi-Tong Ma



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2 p 1 b 1 s 2 2 s 1 2 p 1 1 s 2 2 p 2 c 1 s 2 2 p 2 1 s 2 1 p 2 d 1 s 2 1 p 2

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potassium element has 19 electrons its electron configuration is $1s^2 2s^2 2p^6 3s^2 3p^6 4s^1$ or simply put at $4s^1$ what is the total number of electrons that can be filled in s p and d subshells

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