

Moles and calculations

Define moles: It is the amount of substance which is equal to the Avogadro's number of particles.

Avogadro's number or Avogadro's Constant = 6.02×10^{23} particles

Formulas for calculation of moles:

$$1. \quad \text{mole} = \frac{\text{mass}}{\text{Mass of one mole}}$$

Mass of one mole = molar mass = Mr or Ar

$$2. \quad \text{mole of any gas} = \frac{\text{Volume of gas in cm}^3}{\text{Volume of one mole of gas}} = \frac{\text{Volume of gas in dm}^3}{\text{Volume of one mole of gas}}$$

Volume of one mole of any gas = molar gas volume = 24000 cm^3 or 24 dm^3

$$3. \quad \text{mole of any substance} = \frac{\text{volume in dm}^3 \times \text{concentration in mol dm}^{-3}}{\text{(in the form of solution)}}$$

$$\text{Volume in dm}^3 = \frac{\text{Volume of solution in cm}^3}{1000}$$

$$4. \quad \text{percentage yield} = \frac{\text{Experimental yield}}{\text{Theoretical yield}} \times 100$$

$$5. \quad \text{percentage purity} = \frac{\text{Mass of pure substance}}{\text{Mass of impure substance}} \times 100$$

$$6. \quad \text{percentage composition by mass} = \frac{\text{No. of atoms} \times \text{Ar of atom}}{\text{Mass of total compound}} \times 100$$

$$7. \quad \text{molecular formula} = n \times \text{empirical formula}$$

$$\text{Where } n = \frac{\text{Molecular formula mass}}{\text{Empirical formula mass}}$$

Skill Practice Moles And Formula Practice

Nessar Ahmed



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Chemfile Skills Practice Experiments Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2006 **The Practice of Chemistry** Donald J. Wink, Sharon Fetzner-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book's unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt 832KB **The Practice of Chemistry Study Guide & Solutions Manual** Pamela Mills, Amina El-Ashmawy, 2003-04-14

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