

## Skills Worksheet

## Concept Review Chapter 5: Section 1

## Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

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## **Table of Contents Skills Worksheet Concept Review Earth Chemistry Holt Earth Science**

1. Understanding the eBook Skills Worksheet Concept Review Earth Chemistry Holt Earth Science
  - The Rise of Digital Reading Skills Worksheet Concept Review Earth Chemistry Holt Earth Science
  - Advantages of eBooks Over Traditional Books
2. Identifying Skills Worksheet Concept Review Earth Chemistry Holt Earth Science
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Skills Worksheet Concept Review Earth Chemistry Holt Earth Science
  - User-Friendly Interface
4. Exploring eBook Recommendations from Skills Worksheet Concept Review Earth Chemistry Holt Earth Science
  - Personalized Recommendations
  - Skills Worksheet Concept Review Earth Chemistry Holt Earth Science User Reviews and Ratings
  - Skills Worksheet Concept Review Earth Chemistry Holt Earth Science and Bestseller Lists
5. Accessing Skills Worksheet Concept Review Earth Chemistry Holt Earth Science Free and Paid eBooks
  - Skills Worksheet Concept Review Earth Chemistry Holt Earth Science Public Domain eBooks
  - Skills Worksheet Concept Review Earth Chemistry Holt Earth Science eBook Subscription Services
  - Skills Worksheet Concept Review Earth Chemistry Holt Earth Science Budget-Friendly Options
6. Navigating Skills Worksheet Concept Review Earth Chemistry Holt Earth Science eBook Formats
  - ePub, PDF, MOBI, and More
  - Skills Worksheet Concept Review Earth Chemistry Holt Earth Science Compatibility with Devices
  - Skills Worksheet Concept Review Earth Chemistry Holt Earth Science Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Skills Worksheet Concept Review Earth Chemistry Holt Earth Science
  - Highlighting and Note-Taking Skills Worksheet Concept Review Earth Chemistry Holt Earth Science
  - Interactive Elements Skills Worksheet Concept Review Earth Chemistry Holt Earth Science

8. Staying Engaged with Skills Worksheet Concept Review Earth Chemistry Holt Earth Science
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Skills Worksheet Concept Review Earth Chemistry Holt Earth Science
9. Balancing eBooks and Physical Books Skills Worksheet Concept Review Earth Chemistry Holt Earth Science
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Skills Worksheet Concept Review Earth Chemistry Holt Earth Science
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Skills Worksheet Concept Review Earth Chemistry Holt Earth Science
  - Setting Reading Goals Skills Worksheet Concept Review Earth Chemistry Holt Earth Science
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Skills Worksheet Concept Review Earth Chemistry Holt Earth Science
  - Fact-Checking eBook Content of Skills Worksheet Concept Review Earth Chemistry Holt Earth Science
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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