

The Mole Workbook Questions

W9.1 What is the mass of one mole of:

- (a) lithium atoms _____
- (b) sodium atoms _____
- (c) calcium atoms _____
- (d) iron atoms _____
- (e) silver atoms _____
- (f) lead atoms? _____

W9.2 What is the mass of:

- (a) 10 moles of chlorine atoms _____
- (b) 0.125 mole of bromine molecules _____
- (c) 0.25 mole of H_2O _____
- (d) 0.5 mole of sulfur molecules, S_8 _____
- (e) 0.5 mole of HNO_3 _____
- (f) 0.5 mole of SO_4^{2-} ions _____
- (g) 0.05 mole of oxygen molecules? _____

W9.3 Calculate the mass of:

- (a) 4 moles NaOH _____
- (b) 0.5 mole CaCl_2 _____
- (c) 2 moles Cl_2 _____
- (d) 0.05 mole MgO _____
- (e) 3 moles Fe_2O_3 _____
- (f) 0.1 mole C_3H_8 _____
- (g) 0.6 mole CaCO_3 _____
- (h) 0.3 mole NaCl _____

Skill Practice 31 Moles And Formulas Practice

**Donald J. Wink, Sharon Fetzer-
Gislason, Sheila McNicholas**



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