TEACHER RESOURCE PAGE

- 4. The particles in a gas are more spread out than the particles in a liquid and have less attractive forces acting between them; so, if the same amount of energy is transferred to a gas as to a liquid, the kinetic energy of the particles in a gas will increase more than the kinetic energy of particles in a liquid.
- Convection movement of the hot water as it expands and rises, then cools and contracts, mixes the hot and cold water.
- 6. Students should defend their answers. A heavy cast iron skillet transfers heat evenly, resists temperature changes, and is slow to heat. A thin stainless steel skillet transfers heat quickly, has a high surface temperature directly over the flame, and cools quickly.

SECTION: USING HEAT

- A damp towel has water that absorbs heat energy and evaporates as the air blows through it. Because the air loses heat energy to the water, the air is cooler.
- a. No, the refrigerator releases heat from inside the box out the back of the unit. The refrigerator is in the room, so the heat is still released in the kitchen.
 b. Yes, because shivering causes muscle movements that generate heat
 c. Yes, because the large ears of a jackrabbit act as radiators to radiate heat from the rabbit's blood.
- 3. The two objects that will be at different temperatures, such as an inside wall and an outside wall, should be separated by an air-filled space so that they are not touching, in order to eliminate heat loss by conduction.
- A diesel engine does not use spark plugs. Instead, it compresses the fuelair mixture so much that a spark is ignited.

Math Skills

TEMPERATURE CONVERSION

1. $T_F = 1.8t + 32.0 = (1.8 \times 453) + 32.0$ = 847° F T = t + 273 = 453 + 273 = 726 K

- **2.** $T_F = 1.8t + 32.0 = [1.8(-235)] + 32.0 = -391$ °F
 - T = t + 273 = -235 + 273 = 38 K
- 3. $T_F = 1.8t + 32.0 (1.8 \times 15) + 32.0$ = 59°F
 - T = t + 273 = 15 + 273 = 288 K
- **4.** $T_F = 1.8t + 32.0 = (1.8 \times 190) + 32.0$ T = t + 273 = 190 + 273 = 463 K $= 370^{\circ}\text{F}$
- **5.** $t = \frac{(T_P 32.0)}{1.8} = \frac{(125 32.0)}{1.8} = 52$ °C
 - T = t + 273 = 52 + 273 = 325 K
- **6.** $t = \frac{(T_F 32.0)}{1.8} = \frac{(17 32.0)}{1.8} = -8.3^{\circ}\text{C}$ T = t + 273 = 285 K
 - $t = \frac{(T_F 32.0)}{1.8} = \frac{(83 32.0)}{1.8} = 28$ °C T = t + 273 = 28 + 273 = 301K
- **7.** $t = \frac{(T_F 32.0)}{1.8} = \frac{(-83.0 32.0)}{1.8}$
 - = -64.4°C T = t + 273 = -64.4 + 273 = 209 K
- **8.** $t = \frac{(T_p 32.0)}{1.8} = \frac{(276 32.0)}{1.8}$
 - $-136^{\circ}\mathrm{C}$
- T = t + 273 = 136 + 273 = 409 K
- **9.** t = T 273 = 5100 273 = 4800°C
 - $T_F = 1.8t + 32.0 = 1.8(4800) + 32.0$ = 8700°F
- **10.** $t = T 273 = 1088 273 = 815^{\circ}\text{C}$
 - $T_F = 1.8t + 32.0 = 1.8(815) + 32.0$ = 1.50 × 103°F
- 11. t = T 273 = 3500 273 = 3200°C
 - $T_P = 1.8t \pm 32.0 1.8(3200) \pm 32.0$
- = 5800° F 12. $t = T - 273 = 240 - 273 = -30^{\circ}$ C
 - $T_F = 1.8t + 32.0 = 1.8(-30) 32.0$ = -20° F
- 13. $T_F = 1.8t + 32.0 = 1.8(29) + 32.0 \cdot 84^{\circ}F$ $T_F = 1.8t + 32.0 = 1.8(-11) + 32.0$ $= 12^{\circ}F$
- **14.** $t = T 273 1.0 \times 10^{3} 273$ = 1.0×10^{6} C
- **15.** t = T 273 = 4.25 273 = -269°C
 - $T_F = 1.8t + 32.0 = 1.8(-269) + 32.0$ = -452° F
- 16. a. $\Delta t = \frac{(\Delta T_F)}{1.8} = \frac{(49^{\circ} \text{F})}{1.8} = 27^{\circ} \text{C}$
 - **b.** initial $t = final \ t \Delta t$ = $45^{\circ}F - 49^{\circ}F = -4^{\circ}F$

Skills Math Skills Specific Heat Answers

Cutnell

Skills Math Skills Specific Heat Answers:

Math Skills for the Sciences John G. Pearson, Dennis M. Stone, Richard F. Swindell, 1976 Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction

Science Spectrum Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03

Reading Comprehension and Skills, Grade 6 Carson-Dellosa Publishing, 2014-03-15 Reading Comprehension and Skills for sixth grade is designed to help students develop a strong foundation of reading basics so that they will become competent readers who can advance to more challenging texts It includes engaging passages and stories about a variety of subjects to appeal to all readers. The book also encourages vocabulary development and reinforces reading comprehension through leveled activity pages that target each student s individual needs for support Kelley Wingate s Reading Comprehension and Skills is the perfect choice for both teachers and parents This valuable reading and comprehension skills practice book provides nearly 100 reproducible pages of exciting activities 96 durable flash cards and a motivating award certificate The differentiated activity pages give students the practice they need at a level that is perfect to help them master basic reading comprehension skills necessary to succeed and are great for use at both school and home Holt Chemistry Ralph Thomas Myers,2004 Basic Concepts of Chemistry Leo J. Malone, Theodore O. Dolter, 2011-12-27 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter WileyPLUS sold separately from text Waves, Sound, and Light Glencoe/McGraw-Hill,2001-06 Physics, Volume One: Chapters 1-17 John D. Cutnell, Kenneth W. Johnson, 2014-12-15 Cutnell and Johnson has been the 1 text in the algebra based physics market for almost 20 years The 10th edition brings on new co authors David Young and Shane Stadler both out of LSU The Cutnell offering now includes enhanced features and functionality The authors have been extensively involved in the creation and adaptation of valuable resources for the text This edition includes chapters 1 17 **SAT Subject Test Physics** Kaplan Test Prep, 2017-01-03 Provides an in depth review of concepts covered on the exam test taking strategies a diagnostic tool

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Student Edition McGraw-Hill/Glencoe, 2001-06

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