# **Concept Review**

#### Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10 11 12 13

- ion octet anion cation
- 1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n)

Complete each statement below by writing the correct term or phrase.

- 4. An anion is an ion with a regarive charge.
- 5. Valence electrons occupy the outermost energy level of an atom.
- 7. A cation is an ion with a possitive charge.
- 8. Many stable ions have an electron configuration of a octet

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

else is missing some

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**R Pring** 

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