

2 - 40 HP

**WEST BEND  
OUTBOARD MOTOR  
SERVICE MANUAL  
1956-1960**



# West Bend Elgin Outboard Service Manual 2 40 Hp

**Clifford Lane**



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oxidation  $\text{CO} + \text{CO}_2 + 2\text{e}^-$  reduction  $\text{Fe}^{2+} + 2\text{e}^- \rightarrow \text{Fe}$   $24 \text{ 3 Ag} + \text{Ni}^{3+} \rightarrow 3 \text{ Ag} + \text{Ni}$  oxidation reduction  $25 \text{ Cu}^{2+} + \text{Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$  oxidation reduction  $26 \text{ O}_2$

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 $5\text{CO} + \text{I}_2 \rightarrow 5\text{CO}_2 + \text{I}_2$   $\text{o} \text{ 5 C} + \text{O}_2 \rightarrow 5 \text{ CO}_2$

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web jun 28 2010 answers i determine what is oxidized and what in each reaction identify the oxidizing agent and the reducing agent also  $2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$   $\text{Sr}^0$  to  $\text{Sr}^{2+}$  oxidized reducing agent  $\text{O}^0$  to  $\text{O}^{2-}$  reduced  $\text{Ox Ag} + 2\text{Li} \rightarrow \text{Li}_2\text{Ag}$   $\text{Li}^0$  to  $\text{Li}^{+1}$  oxidized  $\text{red Ag}$   $\text{S}^0$  to  $\text{S}^{2-}$  reduced  $\text{Ox Ag} + 3 \text{ 2Cs} \rightarrow \text{Br}_2$

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