

Name: \_\_\_\_\_ **KEY** \_\_\_\_\_

Recitation Section: \_\_\_\_\_ Group Number: \_\_\_\_\_

## Chemical Equilibrium

### Instructions:

- Listen to an introduction from your recitation instructor.
- Work through the entire activity with your group. Each group member is responsible for contributing equally to this activity, and each person will turn in an assignment following the activity.
- No one is allowed to leave before the recitation period is over. If your group completes the activity early, use the extra time to help other groups.
- The recitation instructors will wrap up the session with an end-of-class 5-minute discussion.
- At the end of the class, one student from each table should gather the individual activity packets and turn them in to the recitation instructor as a set. Each activity packet should be complete.

**Concepts and Skills:**  
**Chemical Equilibrium**  
**The Equilibrium Constant**  
**Le Chatelier's Principle**  
**Buffered Solutions**

**For Further Practice**  
**Chapter 12, Section 12.4**  
**Chapter 12, Section 12.5**  
**Chapter 12, Section 12.6**  
**Chapter 13, Section 13.6**

# Teaching Transparency Chemical Equilibrium Answer Key

**Mark Steyvers**



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**Books in Print** ,1982     Energy Information Abstracts Annual, 1990 ,1990     *The Publishers' Trade List Annual* ,1980

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