The Periodic Table

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Na	Mg											All	Si	P	S	CI	Ar
K	Ca	Sc	Ti	V	Cr	Min	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	To	Ru	Bh	Pd	Ag	Cd	In	Sn	Sb	Te	P)	Xe
Cs	Ва	La	Hf	Ta	W	Re	Os	fr	Pt	Аш	Hg	TI	Pb	Bi	Po	At	Rr
Fr	Ra	Ac	Rf	Ha	Sg	Uns	Uno	Une							344		
			Ce	Pr	Nd	Pm	Sm	Eu	Gd	Тъ	Dy	Но	Er	Tm	Yb	Lu]
			Th	Pa	U	Np	Pu	Am	Cm	Bk	CI	Es	Fm	Md	No	Lr	1

- The elements in periods 2 through 7 that are to the left of the zigzagged line are metals.
 List five metals.
- The elements in periods 1 through 6 that are to the right of the zigzagged line are nonmetals. List five nonmetals.
- The elements that border either side of the zigzagged line are metalloids. Metalloids have properties of both metals and nonmetals. Name two metalloids.
- The alkali metals are in group 1. They have only one electron in their outer shell. They
 are very reactive and have a low melting point. Shade the alkali metals blue.
- The nobel gases are found in group 18. Their outer energy level is filled, and they are very nonreactive, colorless gases. Shade the nobel gases red.
- Halogens have seven electrons in the outer shell. They are located in group 17 on the periodic table. Halogens react with metals to form salts. Shade the halogens green.
- 7. The transition elements are found in groups 3 through 12 and periods 4 through 7. These elements have either one or two electrons in the outer energy level. They are often used to form alloys because they are hard and have high melting points. Shade the transition elements orange.
- Alkaline earth metals are located in group 2. They have two electrons in their outer energy level. Shade the alkaline metals purple.

<u>Teaching Transparency Worksheet The Periodic Table</u> <u>Answer Key</u>

F Rizvi

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Glencoe Science McGraw-Hill Staff, 2001-06 Chemistry in the Community American Chemical Society, 2006-01-31 This laboratory based text centres itself around decision making activities where students apply their chemistry knowledge to realistic situations This fifth edition includes more photographs new drawings and new design Science Spectrum Holt School Library Journal ,1988 Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03 **Periodic Table Cindy** Blobaum, 2005 What do chocolate chip cookies chemistry and logic have in common They are the basis for a unit that lets students become actively engaged in discovering the arrangement of the periodic table This learning activity takes the periodic table out of the static presentation usually associated with textbooks and chemistry courses and interjects an element of discovery The two activities in this unit provide students with information that they have to arrange in organized charts In the process of creating the arrangements students will be involved in problem solving and will gain an appreciation for the scientific process of exploration and verification This dynamic unit meets national science standards in seven teaching and content areas Bring the periodic table to life with this hands on minds on unit Book jacket **Elements: What Are Compounds? Gr. 5-8** George Graybill, 2015-10-01 This is the chapter slice What Are Compounds from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom's Taxonomy and STEM initiatives **Answer Kev for The Mystery of the Periodic Table** Staff of Christ the King Books, 2018-02 The Periodic Table Edward Shevick, 2008-09-01 Implement Newton's First Law of Motion as a teaching principle with this packet students bodies at rest need many hands on activities impressed forces to learn compelling change This collection of Physical Science Action Labs will give your students plenty of experience with matter The labs include determining characteristics of matter focusing specifically on the periodic table The Disappearing Spoon Study Guide Pembroke Notes, 2013-10 How to Use This Book This book is to be used along side the bestselling book The Disappearing Spoon by Sam Kean for anyone who wants to learn about the periodic table in an engaging and unique way For students The study questions are in order and follow Sam Kean's narrative Answer the questions as you read the book The answers are in the back section For teachers This is an easy and interesting resource to help your students learn about the periodic table Never has it been put in a way that transforms a

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Teachers.Net Lesson Exchange: The Periodic Table, Karen Lyn Jenkins presents a lesson intended to familiarize middle school special education students with the periodic table Jenkins includes a list of the materials required the time needed and the procedures Teachers Net provides the lesson as part of the Teachers Net Lesson Exchange online resource

Advancing Scientific Literacy in Chemistry Julia Burton, 2011 The books in this series offer an attractive effective method for developing scientific literacy Contain engaging activities that will draw in even students who feel they cant do science Can be used to introduce or reinforce science concepts and vocabulary Cover a wide range of topics within each broad subject area of biology physics chemistry and earth science

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