

TEACHING TRANSPARENCY WORKSHEET**54****The pH Scale****Use with Chapter 18,
Section 18.3**

1. What is the pH of a solution with a $[\text{H}^+]$ of 10^{-8} ? _____
2. What is the pOH of a solution with a $[\text{OH}^-]$ of 10^{-11} ? _____
3. What is the pH of a solution that has a $[\text{OH}^-]$ of 10^{-2} ? _____
4. What is the pOH of a solution that has a $[\text{H}^+]$ of 10^{-5} ? _____
5. What do you notice about the product of $[\text{H}^+]$ and $[\text{OH}^-]$ for any aqueous solution?

6. What do you notice about the sum of pH and pOH for any aqueous solution?

7. Which is more acidic, a solution with a pH of 6 or one with a pH of 9?

8. Which is more basic, a solution with a pOH of 7 or one with a pOH of 12?

9. Which is more acidic, a solution with a pH of 5 or one with a pOH of 10?

10. Which is more basic, a solution with a pH of 8 or one with a pOH of 12?

11. Stomach contents can have a pH of 3. Are stomach contents acidic, basic, or neutral?

12. Pure water has a pOH of 7. Is pure water acidic, basic, or neutral?

13. Normal rain has a pH of approximately 6. Is normal rain strongly acidic, slightly acidic, neutral, slightly basic, or strongly basic?

14. Acid precipitation is often a problem in industrialized areas. What might you expect the pH of acid rain to be?

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N Noddings



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