

Odyssey Final Exam - Multiple Choice Questions with Correct Answers |2024 Update Graded A+

C. ✓✓ _____ was so angered by Odysseus injuring his son, _____, that he would not allow Odysseus to return home to Ithaca.

- A. Zeus; Hephaestus
- B. Poseidon; Proteus
- C. Poseidon; Polyphemus
- D. Zeus; Helios

B. ✓✓ Which two cities did Telemachus visit on the mainland of Greece?

- A. Mycenae and Athens
- B. Pylos and Sparta
- C. Pylos and Athens
- D. Sparta and Mycenae

B. ✓✓ _____ instructed Menelaus to get home by _____.

- A. Poseidon; making hecatombs
- B. Proteus; first going back to Egypt
- C. Cleopatra; casting a piece of plate into the sea
- D. the Old Man Of The Sea; catching Proteus' beard

D. ✓✓ Which of the following was *NOT* going on in the house of Odysseus while Telemachus was away?

- A. At first the suitors did not know Telemachus had left.
- B. When they realized Telemachus was gone, the suitors plotted his death.
- C. The suitors organized an ambush to kill Telemachus off of Ithaca upon his return.
- D. Penelope organized an archery competition for the suitors to vie for her hand.

D. ✓✓ Calypso's main argument against releasing Odysseus included which of the following?

- A. It was fine for gods to marry mortals.
- B. It was bad for goddesses to marry mortals.
- C. It was fine for gods to marry immortals.
- D. A and B.
- E. A and C.

The Odyssey Semster Final Exam Answers

Kidd, Terry T.



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web this unit focuses on building an initial understanding of the mole and how to use it to quantify chemicals at a reasonable scale students develop a conceptual understanding of the mole and use molar mass and avogadro s number to calculate the number of atoms or molecules in a substance

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web the mole is an amount unit similar to familiar units like pair dozen gross etc it provides a specific measure of the number of atoms or molecules in a sample of matter one latin connotation for the word mole is large mass or bulk which is consistent with its use as the name for this unit

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web libretexts the mole concept is central to any chemistry calculation based on experimental results the mole is how we relate the unbelievably small atoms and molecules that make something up to the measurable properties such as mass which we may observe in a laboratory setting

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web oct 10 2023 mole standard unit $6.02214076 \times 10^{23}$ in chemistry for measuring large quantities of very small entities such as atoms molecules or other specified particles the number of units in a mole also bears the name avogadro s number or avogadro s constant in honor of the italian physicist amedeo avogadro

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web the mole was made the seventh si base unit in 1971 by the 14th cgpm 2019 redefinition of si base units before the 2019 redefinition of the si base units the mole was defined as the amount of substance of a system that contains as many elementary entities as there are atoms in 12 grams of carbon 12 the most common isotope of carbon