

# SOLUBILITY AND TEMPERATURE GIZMO

## SOLUBILITY AND TEMPERATURE

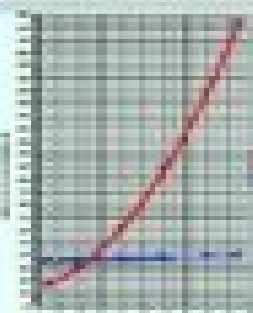
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Potassium Nitrate ( $\text{KNO}_3$ )			
Temperature ( $^{\circ}\text{C}$ )	Solute Added (g)	Solute at Bottom (g)	Solute Dissolved (g)
5	20.00	3.14	16.86
20	40.00	4.35	35.65
40	80.00	18.04	61.96
60	120.00	12.11	107.89
80	180.00	11.41	168.59
100	240.00	16.22	223.78

Sodium Chloride ( $\text{NaCl}$ )			
Temperature ( $^{\circ}\text{C}$ )	Solute Added (g)	Solute at Bottom (g)	Solute Dissolved (g)
5	40.00	4.06	35.94
20	40.00	3.34	36.66
40	40.00	2.39	37.61
60	40.00	1.43	38.57
80	40.00	0.48	39.52
100	60.00	19.76	40.24

## INSIDE:

### SOLUBILITY AND TEMPERATURE



1. Potassium Nitrate ( $\text{KNO}_3$ ) is more soluble than Sodium Chloride ( $\text{NaCl}$ ) at all temperatures.
2. Potassium Nitrate ( $\text{KNO}_3$ ) solubility increases much more rapidly with temperature than Sodium Chloride ( $\text{NaCl}$ ) solubility.
3. At 100°C, Potassium Nitrate ( $\text{KNO}_3$ ) solubility is 223.78 g/100g water, while Sodium Chloride ( $\text{NaCl}$ ) solubility is only 40.24 g/100g water.
4. At 5°C, Potassium Nitrate ( $\text{KNO}_3$ ) solubility is 16.86 g/100g water, while Sodium Chloride ( $\text{NaCl}$ ) solubility is 35.94 g/100g water.
5. The solubility of both substances increases with temperature, but the rate of increase is much greater for Potassium Nitrate.

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**George C. Pimentel, Richard D.  
Spratley**



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