

# The Periodic Table and Periodic Law

## Section 6.1 Development of the Modern Periodic Table

*In your textbook, read about the history of the periodic table's development.*

Use each of the terms below just once to complete the passage.

|          |              |                  |          |
|----------|--------------|------------------|----------|
| octaves  | atomic mass  | atomic number    | nine     |
| elements | properties   | Henry Moseley    | eight    |
| protons  | periodic law | Dmitri Mendeleev | accepted |

The table below was developed by John Newlands and is based on a relationship called the law of (1) \_\_\_\_\_. According to this law, the properties of the elements repeated every (2) \_\_\_\_\_ elements. Thus, for example, element two and element (3) \_\_\_\_\_ have similar properties. The law of octaves did not work for all the known elements and was not generally (4) \_\_\_\_\_.

|   |    |    |    |    |    |    |
|---|----|----|----|----|----|----|
| 1 | 2  | 3  | 4  | 5  | 6  | 7  |
| H | Li | G  | Bo | C  | N  | O  |
| 8 | 9  | 10 | 11 | 12 | 13 | 14 |
| F | Na | Mg | Al | Si | P  | S  |

The first periodic table is mostly credited to (5) \_\_\_\_\_. In his table, the elements were arranged according to increasing (6) \_\_\_\_\_. One important result of this table was that the existence and properties of undiscovered (7) \_\_\_\_\_ could be predicted.

The element in the modern periodic table are arranged according to increasing (8) \_\_\_\_\_, as a result of the work of (9) \_\_\_\_\_. This arrangement is based on number of (10) \_\_\_\_\_ in the nucleus of an atom of the element. The modern form of the periodic table results in the (11) \_\_\_\_\_, which states that when elements are arranged according to increasing atomic number, there is a periodic repetition of their chemical and physical (12) \_\_\_\_\_.

# Teaching Transparency The Periodic Table Answers

**DP Hallahan**



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