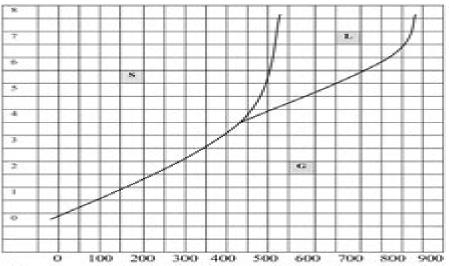
(#1) For each of the questions on this worksheet, refer to the phase diagram for mysterious compound X.



Pressure (atm)
Labels:
S – solid
L- liquid
G - gas

Temperature (degrees Celsius)

(1) If you were to have a bottle containing compound X in your closet, what phase would it most likely be in? _____

(2) At what temperature and pressure will all three phases coexist? _____

(3) If you have a bottle of compound X at a pressure of 3 atm and temperature of 100°C, what will happen if you raise the temperature to 400°C?

(4) Why can't compound X be boiled at a temperature of 200°C?

(5) Is it possible to drink compound X?

(6) What is the critical temperature of compound X?

Teaching Transparency Worksheet Phase Diagrams Answers

Michael Rajnik

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