

4. A solution contains 2.0 g of NaCl in 100 mL of water. The solution is heated to 100°C, and the water is evaporated. How much NaCl is left?
  - a. 2.0 g
  - b. 10.0 g
  - c. 10.0 mL
  - d. 10.0 L
5. A solution contains 2.0 g of NaCl in 100 mL of water. The solution is heated to 100°C, and the water is evaporated. How much NaCl is left?
  - a. 2.0 g
  - b. 10.0 g
  - c. 10.0 mL
  - d. 10.0 L
6. A solution contains 2.0 g of NaCl in 100 mL of water. The solution is heated to 100°C, and the water is evaporated. How much NaCl is left?
  - a. 2.0 g
  - b. 10.0 g
  - c. 10.0 mL
  - d. 10.0 L
7. A solution contains 2.0 g of NaCl in 100 mL of water. The solution is heated to 100°C, and the water is evaporated. How much NaCl is left?
  - a. 2.0 g
  - b. 10.0 g
  - c. 10.0 mL
  - d. 10.0 L
8. A solution contains 2.0 g of NaCl in 100 mL of water. The solution is heated to 100°C, and the water is evaporated. How much NaCl is left?
  - a. 2.0 g
  - b. 10.0 g
  - c. 10.0 mL
  - d. 10.0 L
9. A solution contains 2.0 g of NaCl in 100 mL of water. The solution is heated to 100°C, and the water is evaporated. How much NaCl is left?
  - a. 2.0 g
  - b. 10.0 g
  - c. 10.0 mL
  - d. 10.0 L
10. A solution contains 2.0 g of NaCl in 100 mL of water. The solution is heated to 100°C, and the water is evaporated. How much NaCl is left?
  - a. 2.0 g
  - b. 10.0 g
  - c. 10.0 mL
  - d. 10.0 L

# Standardized Test Practice Chapter 13 Chemistry Answers

**Yan Bai**



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