



### PARTICLE THEORY AND STATES OF MATTER

1 What is particle theory?

2 All of matter is made of particles. Explain in which state of matter the particles are most closely packed?

3 Describe how particles move in a solid. Describe how they move in a liquid and in a gas.

4 The particles in a solid have a regular arrangement. Describe how the particles in a liquid and in a gas are arranged.

5 How are particles in a solid held together?

11 Explain how the particles in a solid are held together. Why do they not move about freely?

Particle theory states that all matter is made up of tiny particles. These particles have three states of matter: solid, liquid and gas. The particles in a solid are packed closely together. In a liquid, the particles are more spread out. In a gas, the particles are very far apart and move freely.

**Solid State:** Particles in a solid are packed closely together. They have a regular arrangement. They are held together by strong forces of attraction. They can only vibrate in their fixed positions.

**Liquid State:** Particles in a liquid are more spread out than in a solid. They can move past each other. They are held together by weaker forces of attraction.

**Gas State:** Particles in a gas are very far apart. They move freely and rapidly in all directions. They are held together by very weak forces of attraction.

**Changes of State:** Particles can change from one state of matter to another. This happens when they gain or lose energy. For example, when a solid is heated, it can melt into a liquid. When a liquid is cooled, it can freeze into a solid.

**Evaporation:** This is the process by which particles in a liquid escape from the surface and become gas. This happens when the particles have enough energy to overcome the forces of attraction.

**Condensation:** This is the process by which particles in a gas lose energy and become a liquid. This happens when the particles do not have enough energy to overcome the forces of attraction.



6 How do the particles in a solid move? Describe how they move in a liquid and in a gas.

10 Describe how the particles in a solid are held together. Why do they not move about freely?

9 What is the difference between a solid and a liquid?

8 Describe how the particles in a solid are arranged. Describe how they are arranged in a liquid and in a gas.

7 How are particles in a solid held together?

Particle Theory and States of Matter Guided Reading

# States Of Matter Guided Reading Answer Key

**R Pring**



## **States Of Matter Guided Reading Answer Key:**

**Properties of Matter: Three States of Matter Gr. 5-8** George Graybill,2015-09-01 This is the chapter slice Three States of Matter from the full lesson plan Properties of Matter Discover what matter is and is not Learn about and the difference between a mixture and a solution Chocked full with hands on activities to understand the various physical and chemical changes to matter Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Written to grade these science concepts are presented in a way that makes them more accessible to students and easier to understand Our resource is jam packed with experiments reading passages and activities all for students in grades 5 to 8 Color mini posters and answer key included and can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives *Activities for a Differentiated Classroom: Level 4* Wendy Conklin,2011-02-01 Easily implement grade appropriate lessons suitable for Grade 4 classrooms Based on current research these easy to use lessons are based on a variety of strategies to differentiate your instruction Activities are included to allow access to all learners ZIP file contains interactive whiteboard compatible resources including sample projects templates and assessment rubrics This resource is correlated to the Common Core State Standards and is aligned to the interdisciplinary themes from the Partnership for 21st Century Skills *Properties of Matter: Mixtures and Solutions Gr. 5-8* George Graybill,2015-09-01 This is the chapter slice Mixtures and Solutions from the full lesson plan Properties of Matter Discover what matter is and is not Learn about and the difference between a mixture and a solution Chocked full with hands on activities to understand the various physical and chemical changes to matter Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Written to grade these science concepts are presented in a way that makes them more accessible to students and easier to understand Our resource is jam packed with experiments reading passages and activities all for students in grades 5 to 8 Color mini posters and answer key included and can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science** ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Properties of Matter: Physical Changes of Matter Gr. 5-8** George Graybill,2015-09-01 This is the chapter slice Physical Changes of Matter from the full lesson plan Properties of Matter Discover what matter is and is not Learn about and the difference between a mixture and a solution Chocked full with hands on activities to understand the various physical and chemical changes to matter Our resource

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