



11 How does the arrangement of particles in a solid differ from the arrangement of particles in a liquid? What happens when a solid melts? What happens when a liquid freezes? What happens when a gas condenses? What happens when a gas expands?

10 How does the arrangement of particles in a solid differ from the arrangement of particles in a liquid? What happens when a solid melts? What happens when a liquid freezes? What happens when a gas condenses? What happens when a gas expands?

9 What is the difference between a solid, a liquid, and a gas?

8 What is the difference between a solid, a liquid, and a gas?

7 What is the difference between a solid, a liquid, and a gas?

PARTICLE THEORY AND STATES OF MATTER

Particle theory explains the properties of matter and the changes of state. It is based on the fact that all matter is made up of tiny particles. These particles have three main properties: they are very small, they are in constant motion, and they have spaces between them. The arrangement and movement of particles explain the different states of matter.

Solids: Particles are packed closely together in a regular arrangement. They vibrate in fixed positions. Solids have a definite shape and volume. **Liquids:** Particles are packed closely together but can move past each other. Liquids have a definite volume but no definite shape. **Gases:** Particles are far apart and move rapidly in all directions. Gases have no definite shape or volume.

Changes of State: When a substance changes from one state to another, its particles change their arrangement and movement. **Melting:** A solid melts when its particles gain enough energy to overcome the forces holding them in a fixed position. **Freezing:** A liquid freezes when its particles lose enough energy to form a fixed arrangement. **Boiling:** A liquid boils when its particles gain enough energy to escape the liquid and become a gas. **Condensing:** A gas condenses when its particles lose enough energy to form a liquid.



Particle Theory and States of Matter
Guided Reading

States Of Matter Guided Reading Answer Key

R Pring



States Of Matter Guided Reading Answer Key:

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