




SCIENCE 8 – STATES OF MATTER WORKSHEET

NAME: _____

Vocabulary			
Contracts	Kinetic molecular theory	Move around quickly	State of matter
Expands	Mass	Slide past each other	Vibrate
Faster	Matter	Slower	Volume

Use your notes from pages 1 – 2 and the terms in the vocabulary box to fill in the blanks for the following nine questions. You will not need to use every term.

- 1) _____ is the amount of matter that makes up something.
- 2) _____ is the amount of space that a material takes up.
- 3) Anything that has mass and volume is called _____.
- 4) When you add energy to matter, the particles move _____ and the matter _____.
- 5) Particles in a solid are packed so close together they can only _____.
- 6) Particles in a liquid can _____.
- 7) Particles in a gas can _____.
- 8) When you remove energy from particles they move _____ and the matter _____.
- 9) The _____ explains how particles act when their spacing and movement change.
- 10) Match each **Term** on the left with the best **Descriptor** on the right. Each **Descriptor** may be used only once.

Term		Descriptor						
	Mass	A.	Anything that has mass and volume					
	Solid	B.	Amount of space an object takes up					
	Gas	C.	Amount of matter in an object					
	Matter	D.		E.		F.		
	Liquid							
	Volume							

States Of Matter Packet Chemistry Answers

Joacim Rocklöv



States Of Matter Packet Chemistry Answers:

Learning Elementary Chemistry for Class 8 (A.Y. 2023-24) Onward Dr. R. Goel, 2023-05-20 The series Learning Elementary Chemistry for Classes 6 to 8 has been revised strictly according to the latest curriculum The content of this series has been developed to fulfill the requirement of all the six domains Concepts Processes Applications Attitudes Creativity and World view of Science to make teaching and learning of Chemistry interesting understandable and enjoyable for young minds This series builds a solid foundation for young learners to prepare them for higher classes The main strength of the series lies in the subject matter and the experience that a learner will get in solving difficult and complex problems of Chemistry Emphasis has been laid upon mastering the fundamental principles of Chemistry rather than specific procedures Unique features of this series are The content of the book is written in a very simple and easy to understand language All the Key concepts in the curriculum have been systematically covered and graded in the text Each theme has been divided into units followed by thought provoking and engaging exercises to test the knowledge understanding and applications of the concepts learnt in that unit At the end of each theme a comprehensive theme assignment which is aligned with the guidelines provided in National Education Policy NEP 2020 is given Explanations illustrations diagrams experiments and solutions to numerical problems have been included to make the subject more interesting comprehensive and appealing Diagrams illustrations and text have been integrated to enhance comprehension Definitions and other important scientific information are highlighted Throughout the series investigations related to the text enable the learners to learn through experimentation Quick revision of each chapter has been given under the caption Highlights in Review Online Support It provides Video lectures Unit wise interactive exercises Chapterwise Worksheet Solution of textbook questions for Teachers only E Book for Teachers only I hope this series would meet the needs and requirements of the curriculum to achieve the learning outcomes as laid down in the curriculum Suggestions and constructive feedback for the further improvement of the book shall be gratefully acknowledged and incorporated in the future edition of the book Author The Science Orbit Chemistry 07 K. Rajalaxmi, Dr. R.L. Madan, The series provides a body of knowledge methods and techniques that characterize science and technology so that students use these efficiently A conscious attempt has been meeting to help students experience science in varied and interesting ways while actively involving them in their own learning **The Nature of Matter Gr. 5-8**,

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The Software Encyclopedia 2000 Bowker Editorial Staff, 2000-05

The Chemical Element Andrew G. Ede, 2006-06-30 One of the most familiar features of any high school chemistry lab is the Periodic Table of Elements Elegant informative useful to any student in the lab the Periodic Table neatly summarizes our scientific knowledge of the chemical elements from hydrogen to uranium and beyond atomic number atomic weight isotopes and more But how did scientists discover all of these features of the elements How did the Periodic Table come to be And even more basically how did the concept of the chemical element come to dominate how scientists understand chemistry This book shows readers the answers to these and other questions regarding the scientific understanding of matter

The Chemical Element a volume in the Greenwood Guides to Great Ideas in Science traces the history of this tremendously powerful concept from the ancient philosophers to the present day The volume covers the idea of the elements held by Aristotle and the other ancient Greek philosophers how Chinese Arabic and other ancient civilizations thought about the elements Mendeleev and the creation of the Periodic Table of Elements the predictive power of which helped in the discovery of dozens of new elements and the discovery of the artificial elements that are heavier than uranium Jargon and mathematics is kept to a minimum and the volumes includes a timeline a glossary and a bibliography making The Chemical Element an ideal resource for students researching chemistry and the history and nature of the scientific understanding of the world around us

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