

1. The table shows some information about solids, liquids and gases.

Complete the table.

state	distance between particles	movement of particles	forces between particles	shape
solid	close together			fixed shape
liquid	close together	move slowly in all directions		shape of container
gas		move quickly in all directions	very weak	

2. This question is about liquids.

(a) When a liquid is put into a beaker it takes on the shape of the beaker.

Why do liquids take on the shape of the beaker?

Tick (✓) the boxes next to **two** correct answers.

Liquids have no fixed shape.

☐

Liquids have no fixed volume.

☐

Liquid particles only vibrate.

☐

The particles in a liquid expand.

☐

The particles in a liquid move around.

☐

(b) Pierre investigates the boiling point of water.

- He heats up some pure water in a beaker until it boils.
- Pierre measures the boiling point of the pure water.
- He does this two more times using new samples of pure water.
- Pierre then does this three more times using salt water.

The table shows some of his results.

type of water	boiling point in °C		
pure water	100	100	100
salt water	102	103	

(i) There is one result missing. The thermometer shows Pierre's missing result.

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