

Stoichiometry Supplemental Practice Problems

1. Butane, C_4H_{10} , the fuel in lighters, burns according to the following equation:

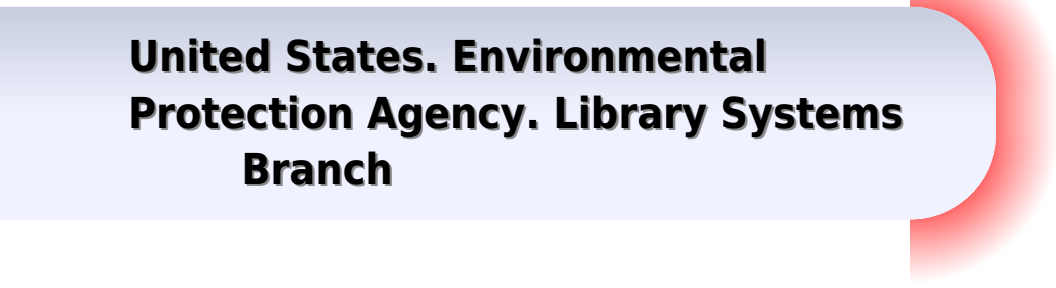


- a. If 3.00 moles of O_2 is to be consumed by this reaction, how many moles of butane will be used up?

0.462 moles C_4H_{10}

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in the resistance to hitler and his anti nazi activities led to his death in a concentration camp