0

THE ARITHMETIC OF EQUATIONS

SECTION REVIEW

Objectives

- Calculate the amount of reactants required or product formed in a non-chemical prisons.
- Interpret halasted chemical equations in terms of interacting region, representative particles, masses, and gas values at STP.

Key Terms

stotch immers.

Part A Completion

Our blue completion exercise to check your wederstanding of the concepts and forms that are introduced in this section. Each blank our by completed with a term, short three, or named on the

The coefficients of a behaviord chemical equation indicate	- N	
the relative number of of reactants and products. All	ž	
statichiomotric calculations begin with a3 Only3	2.	
and4 are conserved in every mastics; moles, volumes,		
and representative particles may not be.	s	
In solinting stotely temperate problems, conversion factors :		
reliable given less of resistants noB of products are used.		
HT - THE GOVERNMENT OF THE STATE HE HAD BEEN AND THE STATE OF THE STAT		

Part B True-False

Classify and of these statem in tries always tray, 57; sometimes true, 57; or mover true, 57.

- The coefficients in a bulanced chemical equation can be used to firm as alle status relating resortants to products.
- B. The coefficients in a trainment chemical equation tell the relative relation of tractants and products, expressed in any switch in unit of values.
- 9. To calculate the mass of a molecule in grams, you can one the molar mass and Aragadre's number

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