

Name _____ Date _____ Class _____

CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY

Covalent Bonding

Section 9.1 The Covalent Bond

In your textbook, read about the nature of covalent bonds.

Use each of the terms below just once to complete the passage.

covalent bond	molecule	sigma bond	exothermic	pi bond
---------------	----------	------------	------------	---------

When sharing of electrons occurs, the attachment between atoms that results is called a(n) **(1) covalent bond**. When such an attachment is formed, bond dissociation energy is released, and the process is **(2) exothermic**. When two or more atoms bond by means of electron sharing, the resulting particle is called a(n) **(3) molecule**. If the electrons shared are contained between the two atoms, the attachment is called a(n) **(4) sigma bond**. If the sharing involves the overlap of parallel orbitals, the attachment is called a(n) **(5) pi bond**.

In your textbook, read about single and multiple bonds and bond strength.

Circle the letter of the choice that best completes the statement or answers the questions.

- In what form do elements such as hydrogen, nitrogen, and oxygen naturally occur?
 - a single atom
 - (b) in molecules containing two atoms**
 - in molecules containing three atoms
 - in molecules containing four atoms
- How many electrons are shared in a double covalent bond?
 - none
 - one
 - two
 - (d) four**
- Bond length is the distance between
 - (a) two molecules of the same substance**
 - the electrons in two attached atoms
 - (c) the nuclei of two attached atoms**
 - the orbitals of two attached atoms
- Which of the following relationships relating to bond length is generally correct?
 - (a) the shorter the bond, the stronger the bond**
 - the shorter the bond, the weaker the bond
 - the shorter the bond, the fewer the electrons in it
 - the shorter the bond, the lower the bond dissociation energy

Study Guide for Content Mastery

Chemistry: Matter and Change • Chapter 9

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Name _____ Date _____ Class _____

CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY

Section 9.2 Naming Molecules

In your textbook, read about how binary compounds and acids are named from their formulas.

For each statement below, write *true* or *false*.

- Binary molecular compounds are generally composed of a metal and a nonmetal. **false**
- The second element in the formula of a binary compound is named using the suffix *-ite*. **false**
- The prefix *mono-* indicates three atoms. **false**
- The prefix *mono-* indicates six atoms. **true**
- In naming the first element in a formula, the prefix *mono-* is not used. **true**
- For binary acids, the hydrogen part of the compound is named using the prefix *hydro-*. **true**
- An oxyacid contains only two elements. **false**
- If the name of the anion of an oxyacid ends in *-ate*, the acid name contains the suffix *-ic*. **false**

In your textbook, read about naming molecular compounds and acids.

For each item in Column A, write the letter of the matching item in Column B.

Column A	Column B
c. N_2O	a. hydrochloric acid
i. NO_2CO	b. dinitrogen tetroxide
g. $\text{H}_2\text{H}_2\text{O}_3$	c. carbon monoxide
e. N_2NH_3	d. nitrous acid
b. $\text{N}_2\text{N}_2\text{O}_3$	e. antimony
d. HNO_3	f. nitric acid
f. HNO_2	g. carbonic acid
a. HBr	h. bromic acid
h. HBrO_3	i. carbon dioxide

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Chemistry: Matter and Change • Chapter 9

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Chapter 1

CH Cherryholmes



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Environmntl Science Clark E. Adams,2004-06 This study guide helps students identify the important concepts from the text and then provides them with review exercises study questions self check exercises and vocabulary review Active Teaching and Learning Strategies Rock Moore,Michelle A. Moore,2004 Active Teaching and Active Learning Strategies Creating a Blueprint for Success is the cumulative result of a year of reflecting asking and listening to questions and

comments that many have concerning the education of our youth The book's genesis however started many years ago as the state of California along with many other states embraced a standards based accountability system This accountability has now been coupled with the recent addition of The No Child Left Behind Legislation In a real sense we are entering both the best and worst of times I say this because of the effects of high stakes testing and the requirement that all students reach high levels of proficiency which admittedly are very lofty goals For this to be realized changes will be required in teaching and assessment as well as a more concentrated focus on student learning The instructional design introduced in this book advocates four components academic social collaborative contextual and cognitive developmental These components work in tandem with Active Teaching and Learning Strategies Combined they form a student responsive instructional methodology which optimizes student success and academic achievement We wanted to create a book that demonstrated to those of us in the field of education the value of looking at academic standards with data about our students and reflecting on how this information should influence our choices concerning instructional practices and our perceptions about the students that we are entrusted with The focus of this endeavor is to see all students achieve and succeed and to keep passion and compassion at the heart of all that we do It is the premise of this book that the difference between effective and outstanding teachers are that outstanding teachers care know how to utilize data practice active teaching strategies and promote the active learning of students Together these characteristics became the design and course of action that Created a Blueprint for Success

Ecology: Teacher's ed, 2005 Human Form, Human Function Kerry L. Hull, 2011 Designed to accompany Human Form Human Function this student workbook offers chapter overviews chapter objectives focus questions mastery tests study activities and mastery test answers **Handbook of Self-Regulation of Learning and Performance** Dale H. Schunk, Jeffrey A. Greene, 2017-09-07 The second edition of the popular Handbook of Self Regulation of Learning and Performance responds to and incorporates the wealth of new research that the first edition inspired on the subject At the same time it advances meaningful perspectives on the scholarship and history that originally shaped the field Divided into five major sections basic domains context technology methodology and assessment and individual and group differences this thoroughly updated handbook addresses recent theoretical refinements and advances in instruction and intervention that have changed approaches to developing learners capabilities to self regulate in educational settings Chapters written by leading experts in the field include discussions of methodological advances and expansions into new technologies and the role of learner differences in such areas as contexts and cultures As a comprehensive guide to a rapidly evolving and increasingly influential subject area this volume represents contemporary and future thinking in self regulation theory research and applications Chapter Structure To ensure uniformity and coherence across chapters each chapter author addresses the theoretical ideas underlying their topic research evidence bearing on these ideas future research directions and implications for educational practice Global A significant number of international contributors are included to reflect the

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