

Name: _____

Atomic Structure & the Periodic Table Study Guide

Atomic Structure

Particle	Charge	Where are they found?
Proton	Positive	Nucleus
Neutron	Neutral/ no charge	Nucleus
Electron	negative	Electron cloud/energy levels outside nucleus

1. What is the rule for writing chemical symbols like Na or Ca? **Capitalize the first letter; lower case second letter**
2. What are the vertical columns on the periodic table called? Groups or families
 - a. How many of these are on the periodic table? 18
 - b. Why are the elements in these columns? They have the same chemical properties (& same # of valence electrons)
 - c. Elements in the same group have the same number of valence electrons.
3. What are the horizontal rows on the periodic table called? periods
 - a. How many of these are on the periodic table? 7
 - b. Elements in these rows have the same number of energy (*electron shells*) levels.
4. What are valence electrons? **Electrons in the outer energy level**
5. How many electrons are in each energy level?
1st energy level 2 2nd energy level 8 3rd energy level 8
6. The atomic number is equal to the # of protons and also # of electrons
7. The atomic mass is equal to the # protons + # of neutrons
8. How do you find the number of neutrons? Atomic mass – atomic number
9. We view the order of the periodic table today according to the atomic number

Study Guide For Periodic Table

Arihant Experts



Study Guide For Periodic Table:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03
The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter. *Hazmat Chemistry Study Guide (Second Edition)* Jill Meryl Levy, 2011 Periodic Table (Advanced) Speedy Publishing, 2014-08-01 An advanced periodic table of elements displays not only the elements but the ions that form each element. A pamphlet with such a visual aid would greatly benefit chemistry students. Any student taking chemistry will need to learn the elements. A pamphlet would be concise and break the information down simply, making it easier to understand and remember. It allows students to simply focus on the main point rather than taking in information that they may or may not need. Know Your 'O' Level Chemistry - A Study Guide,

Understanding Earth Student Study Guide Peter L. Kresan, Reed Mencke, 2006-05-03 The guide helps students prepare for lectures and exams with a heavy emphasis on utilizing the book's Web resources. **Study Guide for The Anatomy and Physiology Learning System** Edith Applegate, 2010-03-30 Designed to accompany The Anatomy and Physiology Learning System 4th Edition by Edith Applegate, this study guide helps you learn and review basic A/P concepts. Each chapter emphasizes medical terminology with a set of key terms, word parts, clinical terms, and abbreviations, and then adds a variety of fun-filled learning exercises, review questions, a quiz, and a word puzzle. The study guide corresponds to the textbook chapter for chapter. Chapter learning objectives help you focus on the most important material. Key concepts are defined on the first page of each chapter in the workbook. Learning exercises for each chapter include short answer, matching, and diagrams to label and color. Self-quizzes allow you to measure your progress and understanding. Fun and Games features end each chapter with a variety of engaging puzzles covering words and concepts. A chapter summary provides a brief review of each chapter. A chapter review provides questions for reinforcement and review of the concepts in each chapter.

Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2021-09-07 provides a complete guide to the fundamentals of chemistry. Page 4 of cover.

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